



STIC Search Report

EIC 1700

STIC Database Tracking Number: 2006100922

TO: Timothy Speer
Location: REM 5D75
Art Unit : 1775
September 8, 2006

Case Serial Number: 10/660578

From: Mei Huang
Location: EIC 1700
REMSEN 4B28
Phone: 571/272-3952
Mei.huang@uspto.gov

Search Notes

Examiner Speer,

Please feel free to contact me if you have any questions or if you would like to refine the search query,

Thank you for using STIC services!

Mei Huang



Banks, Kendra

200922

From: TIMOTHY SPEER [timothy.speer@uspto.gov]
Sent: Thursday, September 07, 2006 8:53 AM
To: STIC-EIC1700
Subject: Database Search Request, Serial Number: 10/660578

Requester:
TIMOTHY SPEER (P/1775)

Art Unit:
GROUP ART UNIT 1775

Employee Number:
70869

Office Location:
REM 05D75

Phone Number:
(571) 272-8385

Mailbox Number:

SCIENTIFIC REFERENCE BR
Sci & Tech Inf. Ctr

SEP 7 2006

Case serial number:
10/660578

Class / Subclass(es):
428/641, 450

Earliest Priority Filing Date:
02/20/03

Format preferred for results:
Paper

Search Topic Information:
I'm looking for the article defined in claims 8-14. It is a silicon substrate which has a patterned metal silicide layer, e.g., a silicide of Ni, Co, Ti, Pt, Fe or Pd, on the silicon substrate and a strain relaxed silicon germanium layer have dislocations on the silicide layer. Key terms include: silicon, metal silicide, patterned, thin line structure, silicon germanium, SiGe, strain relaxed, dislocations, nickel silicide, NiSi, NiSi₂, dislocation density (for the SiGe layer). The article is made by depositing a strain relaxed SiGe layer on a Si substrate, forming a metal layer on the SiGe layer and heating the structure to diffuse the metal of the metal layer through dislocations in the SiGe layer, thereby forming the metal silicide layer at the Si/SiGe interface.

Pat. & T.M. Office

Special Instructions and Other Comments:



STIC Search Results Feedback Form

EIC17000

Questions about the scope or the results of the search? Contact the *EIC searcher or contact:*

**Kathleen Fuller, EIC 1700 Team Leader
571/272-2505 REMSEN 4B28**

Voluntary Results Feedback Form

➤ *I am an examiner in Workgroup:*

Example: 1713

➤ *Relevant prior art found, search results used as follows:*

- 102 rejection
- 103 rejection
- Cited as being of interest.
- Helped examiner better understand the invention.
- Helped examiner better understand the state of the art in their technology.

Types of relevant prior art found:

- Foreign Patent(s)
- Non-Patent Literature
(journal articles, conference proceedings, new product announcements etc.)

➤ *Relevant prior art not found:*

- Results verified the lack of relevant prior art (helped determine patentability).
- Results were not useful in determining patentability or understanding the invention.

Comments:

Drop off or send completed forms to EIC1700 REMSEN 4B28

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DICTIONARY FILE UPDATES: 7 SEP 2006 HIGHEST RN 906063-52-3

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FILE LAST UPDATED: 7 Sep 2006 (20060907/ED)

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(FILE 'HOME' ENTERED AT 15:54:33 ON 08 SEP 2006)

FILE 'HCAPLUS' ENTERED AT 15:55:13 ON 08 SEP 2006
L1 1 SEA US2004166329/PN

FILE 'REGISTRY' ENTERED AT 15:57:14 ON 08 SEP 2006
L2 2 SEA (12727-59-2/BI OR 7440-21-3/BI)

L3 1 SEA 7440-21-3/RN
 L4 770 SEA (SI(L)GE)/ELS (L) 2/ELC.SUB
 L5 1 SEA L2 AND L4
 L6 1978 SEA (SI(L)(NI OR CO OR TI OR PT OR FE OR PD))/ELS (L)
 2/ELC.SUB

FILE 'HCAPLUS' ENTERED AT 16:17:06 ON 08 SEP 2006
 L7 499731 SEA L3 OR (SILICON OR SI) (2A) (SUBSTRAT? OR SURFACE? OR
 BASE# OR SUBSTRUCT?)
 L8 17736 SEA L4
 L9 1219 SEA L5
 L10 33495 SEA L6
 L11 4602 SEA (SILICON(A)GERMANIUM OR SI(A)GE) (3A) (FILM? OR
 THINFILM? OR LAYER?)
 L12 13462 SEA L7 AND (L8 OR L11) Si and Si:Ge
 L13 150268 SEA METAL? (3A) (FILM? OR THINFILM? OR LAYER?)
 L14 723 SEA L12 AND (L10 OR L13)
 L15 QUE MULTILAYER? OR (MULTI? OR PLURAL? OR SEVERAL) (W) LAYER
 ?
 L16 36 SEA L14 AND L15
 L17 QUE PATTERN?
 L18 64 SEA L14 AND L17
 L19 1 SEA L16 AND L18
 L20 QUE DISLOCAT?
 L21 21 SEA L14 AND L20
 L22 1 SEA L18 AND L21
 L23 QUE PENETRAT? OR SATURAT?
 L24 QUE DIFFUS? OR SPREAD? OR DISTRIBUT?
 L25 10 SEA L14 AND L23
 L26 145 SEA L14 AND L24
 L27 3 SEA L25 AND L26
 L28 5 SEA L19 OR L22 OR L27
 L29 34 SEA L16 NOT L28

=> d l28 ibib abs hitstr hitind 1-5

L28 ANSWER 1 OF 5 HCAPLUS COPYRIGHT 2006 ACS on STN
 ACCESSION NUMBER: 2005:728284 HCAPLUS
 DOCUMENT NUMBER: 143:357432
 TITLE: Formation of nickel silicide layer on
 strained-Si0.83Ge0.17/Si(001) using a
 sacrificial Si layer and its morphological
 instability
 AUTHOR(S): Jang, Chi Hwan; Shin, Dong Ok; Baik, Sung Il;
 Kim, Young-Woon; Song, Young-Joo; Shim,
 Kyu-Hwan; Lee, Nae-Eung
 CORPORATE SOURCE: Department of Materials Engineering and Center
 for Advanced Plasma Surface Technology,
 Sungkyunkwan University, Kyunggi-do, 440-746, S.
 Korea
 SOURCE: Japanese Journal of Applied Physics, Part 1:
 Regular Papers, Brief Communications & Review
 Papers (2005), 44(7A), 4805-4813
 CODEN: JAPNDE
 PUBLISHER: Japan Society of Applied Physics
 DOCUMENT TYPE: Journal
 LANGUAGE: English
 AB Nickel silicide was formed on strained-Si0.83Ge0.17/Si(001) using a

sacrificial Si capping (cap-Si) layer and its morphol. characteristics were investigated. Nickel silicide layers were grown by rapid thermal annealing of the samples with the structure of Ni (.simeq. 14 nm)/cap-Si (.simeq. 26 nm)/Si0.83Ge0.17/Si(001) at the annealing temp. (TA) range of 400-800°C. The phase formation, surface and interfacial morphologies, and elec. properties of the resulting samples were characterized by various measurement techniques, including X-ray diffraction, at. force microscopy, SEM, Auger electron spectroscopy, cross-sectional transmission electron microscopy, and the four-point probe method. The results showed the formation of a uniform layer nickel monosilicide (NiSi) with a thickness of .simeq.30 nm at 400-550°C and sheet resistance values of 6.5-7.9 Ω/box.. The sheet resistance values of the samples annealed at TA ≥ 600°C were found to be increased, however, and this is attributed to the agglomeration of nickel monosilicide leading to discrete large-size NiSi grains. Microstructural and chem. analyses of the samples annealed at elevated temp., TA ≥ 750°C, indicated the formation of large agglomerated NiSi grains penetrating into the Si0.83Ge0.17/Si(001) structure and the conversion of the cap-Si layer situated in between the nickel silicide grains into an Si1-uGeu layer (u .simeq. 0.01-0.03), due to the out-diffusion of Ge from the SiGe layer during agglomeration. However, no NiSi2 phase was obsd. at these elevated annealing temps.

IT 12035-57-3, Nickel silicide
 RL: FMU (Formation, unclassified); PRP (Properties); FORM (Formation, nonpreparative)
 (formation of nickel silicide layer on strained-Si0.83Ge0.17/Si(001) using a sacrificial Si layer and its morphol. instability)
 RN 12035-57-3 HCAPLUS
 CN Nickel silicide (NiSi) (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)

Ni≡Si

IT 7440-21-3, Silicon, properties 113677-38-6, Germanium 17, silicon 83 (atomic)
 RL: PRP (Properties)
 (formation of nickel silicide layer on strained-Si0.83Ge0.17/Si(001) using a sacrificial Si layer and its morphol. instability)
 RN 7440-21-3 HCAPLUS
 CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

RN 113677-38-6 HCAPLUS
 CN Silicon alloy, base, Si 65,Ge 35 (9CI) (CA INDEX NAME)

Component	Component	Component
Percent	Registry Number	
Si	65	7440-21-3
Ge	35	7440-56-4

CC 76-2 (Electric Phenomena)
 IT Diffusion
 (out-diffusion; formation of nickel silicide layer on
 strained-Si0.83Ge0.17/Si(001) using a sacrificial Si layer and
 its morphol. instability)
 IT 12035-57-3, Nickel silicide
 RL: FMU (Formation, unclassified); PRP (Properties); FORM
 (Formation, nonpreparative)
 (formation of nickel silicide layer on strained-
 Si0.83Ge0.17/Si(001) using a sacrificial Si layer and its
 morphol. instability)
 IT 7440-21-3, Silicon, properties 113677-38-6,
 Germanium 17, silicon 83 (atomic)
 RL: PRP (Properties)
 (formation of nickel silicide layer on strained-
 Si0.83Ge0.17/Si(001) using a sacrificial Si layer and its
 morphol. instability)
 REFERENCE COUNT: 36 THERE ARE 36 CITED REFERENCES AVAILABLE
 FOR THIS RECORD. ALL CITATIONS AVAILABLE
 IN THE RE FORMAT

L28 ANSWER 2 OF 5 HCAPLUS COPYRIGHT 2006 ACS on STN
 ACCESSION NUMBER: 2004:701646 HCAPLUS
 DOCUMENT NUMBER: 141:216852
 TITLE: Method for fabricating a thin line structure,
 multilayered structure, and multilayered
 intermediate structure
 INVENTOR(S): Sakai, Akira; Zaima, Shigeaki; Yasuda, Yukio;
 Nakatsuka, Osamu
 PATENT ASSIGNEE(S): Nagoya University, Japan
 SOURCE: U.S. Pat. Appl. Publ., 7 pp.
 CODEN: USXXCO
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 2004166329	A1	20040826	US 2003-660578	200309 12
JP 2004253607	A2	20040909	JP 2003-42275	200302 20
PRIORITY APPLN. INFO.:			JP 2003-42275	A 200302 20

AB On a given silicon substrate is epitaxially
 grown a strain-relaxed silicon germanium
 layer with penetrated dislocations and formed a
 metallic layer to form a multilayered intermediate
 structure, which is heated. In this case, metallic elements of the
 metallic layer are diffused through the
 penetrated dislocations of the silicon
 germanium layer to form a thin line structure made
 of metallic silicide at a boundary face between the silicon
 base and the silicon germanium

layer.
 IT 7440-21-3, Silicon, uses 12727-59-2
 RL: DEV (Device component use); USES (Uses)
 (method for fabrication of a thin line and multilayered
 intermediate structure)
 RN 7440-21-3 HCPLUS
 CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

RN 12727-59-2 HCPLUS
 CN Germanium alloy, base, Ge 0-100, Si 0-100 (9CI) (CA INDEX NAME)

Component	Component Percent	Component Registry Number
Ge	0 - 100	7440-56-4
Si	0 - 100	7440-21-3

IC ICM B32B015-00
 INCL 428432000; 428433000
 CC 76-3 (Electric Phenomena)
 IT 7440-21-3, Silicon, uses 12727-59-2
 RL: DEV (Device component use); USES (Uses)
 (method for fabrication of a thin line and multilayered
 intermediate structure)

L28 ANSWER 3 OF 5 HCPLUS COPYRIGHT 2006 ACS on STN
 ACCESSION NUMBER: 2004:21042 HCPLUS
 DOCUMENT NUMBER: 140:69109
 TITLE: Semiconductor device and its manufacturing
 method
 INVENTOR(S): Watanabe, Heiji; Endo, Kazuhiko; Manabe, Kenzo
 PATENT ASSIGNEE(S): NEC Corporation, Japan
 SOURCE: PCT Int. Appl., 41 pp.
 CODEN: PIXXD2
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
-----	-----	-----	-----	-----
WO 2004004014	A1	20040108	WO 2003-JP7761	200306 19

W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH,
 CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD,
 GE, GH, GM, HR, HU, ID, IL, IN, IS, KE, KG, KP, KR, KZ, LC,
 LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NI,
 NO, NZ, OM, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, TJ,
 TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW
 RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ,
 BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK,
 EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PT, RO, SE,
 SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR,
 NE, SN, TD, TG

JP 2004031760	A2	20040129	JP 2002-187596	200206
				27
AU 2003244275	A1	20040119	AU 2003-244275	200306
				19
CN 1663051	A	20050831	CN 2003-815014	200306
				19
US 2005247985	A1	20051110	US 2004-519084	200412
				23
PRIORITY APPLN. INFO.:				JP 2002-187596 A
				200206
				27
			WO 2003-JP7761	W
				200306
				19

AB A semiconductor device having a gate insulating film and a gate electrode formed in this order on a **Si substrate** is described, where the gate insulating film includes a N-contg. high dielec. const. insulating film with a structure in which N is introduced into a metal oxide or a metal silicate, the N concn. in the N-contg. high dielec. const. insulating film has a **distribution** in the direction of the film thickness, and the position where the N concn. is max. in the direction of the film thickness is present in a region away from the **Si substrate**. A method for manufg. a semiconductor device comprising a step of introducing N into a high dielec. const. insulating film of a metal oxide or a metal silicate by exposure to a N-contg. plasma is also disclosed. As a result, the thermal stability of the high dielec. const. gate insulating film is improved, and the dopant **penetration** is suppressed, thereby preventing the elec. characteristics of the interface with the **Si substrate** from degrading.

IT 7440-21-3, Silicon, uses 11148-21-3

RL: DEV (Device component use); USES (Uses)
(N-contg. gate insulator of semiconductor device and its manufg. method)

RN 7440-21-3 HCAPLUS

CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

RN 11148-21-3 HCAPLUS
CN Germanium alloy, nonbase, Ge,Si (9CI) (CA INDEX NAME)

Component	Component
Registry Number	
Ge	7440-56-4
Si	7440-21-3

IC ICM H01L029-78

ICS H01L021-336

CC 76-3 (Electric Phenomena)

IT 1314-23-4, Zirconia, uses 1344-28-1, Alumina, uses
 7440-21-3, Silicon, uses 7631-86-9, Silica, uses
 11148-21-3 12033-89-5, Silicon nitride, uses 37248-04-7,
 Hafnium silicate
 RL: DEV (Device component use); USES (Uses)
 (N-contg. gate insulator of semiconductor device and its manufg.
 method)

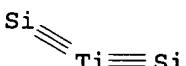
REFERENCE COUNT: 13 THERE ARE 13 CITED REFERENCES AVAILABLE
 FOR THIS RECORD. ALL CITATIONS AVAILABLE
 IN THE RE FORMAT

L28 ANSWER 4 OF 5 HCAPLUS COPYRIGHT 2006 ACS on STN
 ACCESSION NUMBER: 1998:727501 HCAPLUS
 DOCUMENT NUMBER: 130:9399
 TITLE: A 42-GHz (fmax) SiGe-base HBT using reduced
 pressure CVD
 AUTHOR(S): Cho, D.-H.; Ryum, B. R.; Han, T.-H.; Lee, S.-M.;
 Shin, S.-C.; Lee, C.
 CORPORATE SOURCE: Semiconductor Technology Division, Electronics
 and Telecommunications Research Institute,
 Taejon, 305-600, S. Korea
 SOURCE: Solid-State Electronics (1998), 42(9), 1641-1649
 CODEN: SSELAS; ISSN: 0038-1101
 PUBLISHER: Elsevier Science Ltd.
 DOCUMENT TYPE: Journal
 LANGUAGE: English

AB A SiGe HBT having a fmax higher than fT was fabricated using a
 prodn. CVD reactor which allows SiH₂Cl₂-based Si
 collector epi-growth at high rate as well as SiH₄-based SiGe base
 epi-growth at low rate. Transistor design together with process
 integration was focused on lowering the extrinsic base resistance
 and the collector-base capacitance. To this purpose, a TiSi₂ layer
 with a sheet resistance of 1.3 Ω/sq was used as a base
 electrode and a selectively implanted collector was used. For the
 base layer, an undoped-Si (300 Å)/p-SiGe (200 Å, NA = 4.4
<sup>+ 10¹⁸ cm⁻³, linearly-graded Ge compn. from 0 to
 0.19)/undoped-Si_{0.81}Ge_{0.19} (110 Å)/undoped-Si (300 Å)
 multilayer was deposited on a LOCOS-patterned
 wafer. To form the emitter-base junction and to activate the As
 dopants in the polysilicon-emitter, rapid thermal annealing (RTA) at
 900° for 20 s was performed only one time so that
 outdiffusion of the B in the base could be suppressed. The
 collector and base currents are nearly ideal. The authors obtained
 a fT of 37 GHz which is near the theor. limit imposed by BVCEO and a
 fmax of 42 GHz. The base resistance and the collector-base
 capacitance extd. from measured S-parameters have a value of 37
 Ω and 27.2 fF, resp.</sup>

IT 12039-83-7, Titanium silicide (TiSi₂)
 RL: DEV (Device component use); PEP (Physical, engineering or
 chemical process); PROC (Process); USES (Uses)
 (base electrode; fabrication of silicon-germanium
 base HBT using reduced-pressure CVD)

RN 12039-83-7 HCAPLUS
 CN Titanium silicide (TiSi₂) (6CI, 8CI, 9CI) (CA INDEX NAME)



IT 118392-03-3, Titanium silicide (TiSi2.6)
 RL: DEV (Device component use); PEP (Physical, engineering or
 chemical process); PROC (Process); USES (Uses)
 (electrodes; fabrication of silicon-germanium
 base HBT using reduced-pressure CVD)

RN 118392-03-3 HCAPLUS

CN Titanium silicide (TiSi2.6) (9CI) (CA INDEX NAME)

Component	Ratio	Component Registry Number
Ti	1	7440-32-6
Si	2.6	7440-21-3

IT 7440-21-3P, Silicon, processes 37380-03-3P,
 Germanium 20, silicon 80 (atomic) 83590-41-4P, Germanium
 0-19, silicon 81-100 (atomic) 115675-33-7P, Germanium 19,
 silicon 81 (atomic)
 RL: DEV (Device component use); PEP (Physical, engineering or
 chemical process); SPN (Synthetic preparation); PREP (Preparation);
 PROC (Process); USES (Uses)
 (fabrication of silicon-germanium base HBT
 using reduced-pressure CVD)

RN 7440-21-3 HCAPLUS

CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

RN 37380-03-3 HCAPLUS
 CN Silicon alloy, base, Si 61,Ge 39 (9CI) (CA INDEX NAME)

Component	Component Percent	Component Registry Number
Si	61	7440-21-3
Ge	39	7440-56-4

RN 83590-41-4 HCAPLUS

CN Silicon alloy, base, Si 62-100,Ge 0-38 (9CI) (CA INDEX NAME)

Component	Component Percent	Component Registry Number
Si	62 - 100	7440-21-3
Ge	0 - 38	7440-56-4

RN 115675-33-7 HCAPLUS

CN Silicon alloy, base, Si 62,Ge 38 (9CI) (CA INDEX NAME)

Component	Component Percent	Component Registry Number
Si	62	7440-21-3
Ge	38	7440-56-4

CC 76-3 (Electric Phenomena)

Section cross-reference(s): 75

ST fabrication silicon germanium base HBT; reduced

IT pressure CVD germanium silicon
 Electric resistance
 (base; from fabrication of silicon-germanium
 base HBT using reduced-pressure CVD)
 IT Vapor deposition process
 (chem.; fabrication of silicon-germanium base
 HBT using reduced-pressure CVD)
 IT Electric capacitance
 (collector-base; from fabrication of silicon-germanium
 base HBT using reduced-pressure CVD)
 IT Heterojunction bipolar transistors
 Rapid thermal annealing
 Semiconductor device fabrication
 (fabrication of silicon-germanium base HBT
 using reduced-pressure CVD)
 IT Diffusion
 (out-diffusion; fabrication of silicon-germanium
 base HBT using reduced-pressure CVD)
 IT 12039-83-7, Titanium silicide (TiSi2)
 RL: DEV (Device component use); PEP (Physical, engineering or
 chemical process); PROC (Process); USES (Uses)
 (base electrode; fabrication of silicon-germanium
 base HBT using reduced-pressure CVD)
 IT 118392-03-3, Titanium silicide (TiSi2.6)
 RL: DEV (Device component use); PEP (Physical, engineering or
 chemical process); PROC (Process); USES (Uses)
 (electrodes; fabrication of silicon-germanium
 base HBT using reduced-pressure CVD)
 IT 7440-21-3P, Silicon, processes 37380-03-3P,
 Germanium 20, silicon 80 (atomic) 83590-41-4P, Germanium
 0-19, silicon 81-100 (atomic) 115675-33-7P, Germanium 19,
 silicon 81 (atomic)
 RL: DEV (Device component use); PEP (Physical, engineering or
 chemical process); SPN (Synthetic preparation); PREP (Preparation);
 PROC (Process); USES (Uses)
 (fabrication of silicon-germanium base HBT
 using reduced-pressure CVD)
 IT 7440-38-2, Arsenic, uses
 RL: MOA (Modifier or additive use); USES (Uses)
 (fabrication of silicon-germanium base HBT
 using reduced-pressure CVD)
 IT 7440-42-8, Boron, processes
 RL: MOA (Modifier or additive use); PEP (Physical, engineering or
 chemical process); PROC (Process); USES (Uses)
 (fabrication of silicon-germanium base HBT
 using reduced-pressure CVD)
 IT 4109-96-0, Dichlorosilane 7803-62-5, Silicon hydride (SiH4), uses
 RL: NUU (Other use, unclassified); USES (Uses)
 (precursor; fabrication of silicon-germanium
 base HBT using reduced-pressure CVD)

REFERENCE COUNT: 23 THERE ARE 23 CITED REFERENCES AVAILABLE
 FOR THIS RECORD. ALL CITATIONS AVAILABLE
 IN THE RE FORMAT

L28 ANSWER 5 OF 5 HCPLUS COPYRIGHT 2006 ACS on STN
 ACCESSION NUMBER: 1992:582595 HCPLUS
 DOCUMENT NUMBER: 117:182595
 TITLE: In situ transmission electron microscopy
 measurements of the electrical and structural
 properties of strained layer germanium

AUTHOR(S): silicide/silicon p-n junctions
 Ross, F. M.; Hull, R.; Bahnck, D.; Bean, J. C.;
 Peticolas, L. J.; Hamm, R. A.; Huggins, H. A.
 CORPORATE SOURCE: Bell Lab., AT and T, Murray Hill, NJ, 07974, USA
 SOURCE: Journal of Vacuum Science & Technology, B:
 Microelectronics and Nanometer Structures
 (1992), 10(4), 2008-12
 DOCUMENT TYPE: CODEN: JVTBD9; ISSN: 0734-211X
 Journal
 LANGUAGE: English

AB The correlation between the **dislocation** structure and the elec. properties of strained layer p-n junction diodes was studied by examg. both structural and elec. characteristics simultaneously in a transmission electron microscope. Device characteristics and structural changes are followed on a single specimen as it is subjected in situ to heat treatment designed to induce relaxation. The nucleation and growth of **dislocations** at the strained interfaces are described for different diode geometries and the influence of these **dislocations** on the elec. properties of the diodes. The results indicate the dominant role of nucleation sites in the relaxation process. An enhancement in layer stability after patterning and **metallizing** strained layer heterostructures was obsd.

IT 7440-21-3, Silicon, uses
 RL: PRP (Properties)
 (semiconductor p-n junction contg. germanium silicide and, elec. and structural properties of)
 RN 7440-21-3 HCPLUS
 CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

IT 37232-85-2 37380-03-3
 RL: PRP (Properties)
 (semiconductor p-n junction contg. silicon and, elec. and structural properties of)
 RN 37232-85-2 HCPLUS
 CN Silicon alloy, base, Si 69,Ge 31 (9CI) (CA INDEX NAME)

Component	Component	Component
Percent	Registry Number	
Si	69	7440-21-3
Ge	31	7440-56-4

RN 37380-03-3 HCPLUS
 CN Silicon alloy, base, Si 61,Ge 39 (9CI) (CA INDEX NAME)

Component	Component	Component
Percent	Registry Number	
Si	61	7440-21-3
Ge	39	7440-56-4

CC 76-3 (Electric Phenomena)
 IT 7440-21-3, Silicon, uses
 RL: PRP (Properties)
 (semiconductor p-n junction contg. germanium silicide and, elec.

and structural properties of)
 IT 37232-85-2 37380-03-3
 RL: PRP (Properties)
 (semiconductor p-n junction contg. silicon and, elec. and
 structural properties of)

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L29 ANSWER 1 OF 34 HCAPLUS COPYRIGHT 2006 ACS on STN
 ACCESSION NUMBER: 2006:37001 HCAPLUS
 DOCUMENT NUMBER: 144:119530
 TITLE: Silicon nitride film with stress control is
 semiconductor device
 INVENTOR(S): Iyer, R. Suryanarayanan; Lam, Andrew M.; Maeda,
 Yuji; Mele, Thomas; Nouri, Faran; Smith, Jacob
 W.; Seutter, Sean M.; Tandon, Sanjeev; Singh
 Thakur, Randhir P.; Thirupapuliyur, Sunderraj
 PATENT ASSIGNEE(S): USA
 SOURCE: U.S. Pat. Appl. Publ., 26 pp.
 CODEN: USXXCO
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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US 2006009041	A1	20060112	US 2004-885969	200407 06

WO 2006014471	A1	20060209	WO 2005-US23933	200507 05

W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA,
 CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI,
 GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM,
 KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN,
 MW, MX, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU,
 SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT, TZ, UA,
 UG, US, UZ, VC, VN, YU, ZA, ZM, ZW
 RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU,
 IE, IS, IT, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR,
 BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD,
 TG, BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM,
 ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM

PRIORITY APPLN. INFO.: US 2004-885969 A
 200407
06

AB The embodiments of the invention pertain to methods for forming a
 nitride etch stop film and a **multilayer** nitride etch stop
 stack to mech. create a controlled stress (tensile or compressive)
 in a semiconductor device. An assembly comprises a
multilayer nitride stack having nitride etch stop layers
 formed on top of one another, each of the nitride etch stop layers
 is formed using a film forming process. A method of making the
multilayer nitride stack includes placing a substrate in a
 single wafer deposition chamber and thermally shocking the substrate

momentarily prior to deposition. A first nitride etch stop layer is deposited over the substrate. A second nitride etch stop layer is deposited over the first nitride etch stop layer.

IT 7440-21-3, Silicon, uses 12790-21-5
 RL: DEV (Device component use); USES (Uses)
 (silicon nitride film with stress control)
 RN 7440-21-3 HCPLUS
 CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

RN 12790-21-5 HCPLUS
 CN Silicon alloy, base, Si,Ge (9CI) (CA INDEX NAME)

Component	Component
Registry Number	
Si	7440-21-3
Ge	7440-56-4

IT 11104-62-4, Cobalt silicide 39467-10-2, Nickel silicide
 RL: DEV (Device component use); PRP (Properties); USES (Uses)
 (silicon nitride film with stress control)
 RN 11104-62-4 HCPLUS
 CN Cobalt silicide (9CI) (CA INDEX NAME)

Component	Ratio	Component Registry Number
Co	x	7440-48-4
Si	x	7440-21-3

RN 39467-10-2 HCPLUS
 CN Nickel silicide (9CI) (CA INDEX NAME)

Component	Ratio	Component Registry Number
Si	x	7440-21-3
Ni	x	7440-02-0

INCL 438724000
 CC 76-3 (Electric Phenomena)
 IT 7440-21-3, Silicon, uses 7631-86-9, Silicon oxide, uses 12790-21-5
 RL: DEV (Device component use); USES (Uses)
 (silicon nitride film with stress control)
 IT 11104-62-4, Cobalt silicide 39467-10-2, Nickel silicide
 RL: DEV (Device component use); PRP (Properties); USES (Uses)
 (silicon nitride film with stress control)

L29 ANSWER 2 OF 34 HCPLUS COPYRIGHT 2006 ACS on STN
 ACCESSION NUMBER: 2005:1314023 HCPLUS
 DOCUMENT NUMBER: 144:62731
 TITLE: Fabrication of single-metal gate material CMOS
 using strained Si-silicon

**germanium heterojunction layered
substrate**

INVENTOR(S): Antoniadis, Dimitri A.; Hoyt, Judy L.; Jung, Jongwan; Yu, Shaofeng
PATENT ASSIGNEE(S): Massachusetts Institute of Technology, USA
SOURCE: PCT Int. Appl., 24 pp.
CODEN: PIXXD2
DOCUMENT TYPE: Patent
LANGUAGE: English
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE		
-----	-----	-----	-----	-----		
WO 2005119762	A1	20051215	WO 2005-US18514	200505 26		
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW	RW: BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG	US 2005274978	A1	20051215	US 2005-138951	200505 26
PRIORITY APPLN. INFO.:			US 2004-575039P	P	200405 27	

AB A strained Si/strained SiGe dual-channel layer substrate provides mobility advantage and when used as CMOS substrate enables single work-function metal-gate electrode technol. A single metal electrode with work-function of 4.5 eV produces near ideal CMOS performance on a dual-channel layer substrate that consists sequentially of a silicon wafer, an epitaxially grown Si0.7Ge0.3 relaxed layer, a compressively strained Si0.4Ge0.6 layer, and a tensile-strained Si cap layer.

IT 7440-21-3, Silicon, uses 11148-22-4
12623-04-0 12675-06-8

RL: DEV (Device component use); USES (Uses)
(layer; fabrication of single-metal gate
material CMOS using strained Si-silicon
germanium heterojunction layered substrate)

RN 7440-21-3 HCAPLUS

CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

RN 11148-22-4 HCAPLUS

CN Germanium alloy, base, Ge,Si (9CI) (CA INDEX NAME)

Component	Component
Registry Number	
Ge	7440-56-4
Si	7440-21-3

RN 12623-04-0 HCAPLUS
 CN Germanium alloy, base, Ge 53, Si 47 (9CI) (CA INDEX NAME)

Component	Component	Component
Percent	Registry Number	
Ge	53	7440-56-4
Si	47	7440-21-3

RN 12675-06-8 HCAPLUS
 CN Germanium alloy, base, Ge 79, Si 21 (9CI) (CA INDEX NAME)

Component	Component	Component
Percent	Registry Number	
Ge	79	7440-56-4
Si	21	7440-21-3

IC ICM H01L021-8238
 CC 76-3 (Electric Phenomena)
 Section cross-reference(s): 56
 ST gate electrode CMOS transistor silicon germanium heterojunction
 multilayer
 IT MOS devices
 (complementary; fabrication of single-metal gate material CMOS
 using strained Si-silicon germanium
 heterojunction layered substrate)
 IT Gate contacts
 (electrode; fabrication of single-metal gate material CMOS using
 strained Si-silicon germanium heterojunction
 layered substrate)
 IT MOSFET (transistors)
 Semiconductor device fabrication
 Semiconductor heterojunctions
 (fabrication of single-metal gate material CMOS using strained
 Si-silicon germanium heterojunction
 layered substrate)
 IT 7440-21-3, Silicon, uses 11148-22-4
 12623-04-0 12675-06-8
 RL: DEV (Device component use); USES (Uses)
 (layer; fabrication of single-metal gate
 material CMOS using strained Si-silicon
 germanium heterojunction layered substrate)
 IT 25583-20-4, Titanium nitride (TiN)
 RL: DEV (Device component use); USES (Uses)
 (single metal gate electrode; fabrication of single-metal gate
 material CMOS using strained Si-silicon
 germanium heterojunction layered substrate)

REFERENCE COUNT: 8 THERE ARE 8 CITED REFERENCES AVAILABLE FOR
 THIS RECORD. ALL CITATIONS AVAILABLE IN
 THE RE FORMAT

ACCESSION NUMBER: 2005:301655 HCAPLUS
 DOCUMENT NUMBER: 142:383683
 TITLE: Fabrication of a multilayer structure
 with an exposed metal layer
 INVENTOR(S): Ruttkowski, Eike; Ilcali, Guerkan; Luyken, R.
 Johannes; Hofmann, Franz; Alba, Manuela
 PATENT ASSIGNEE(S): Infineon Technologies A.-G., Germany
 SOURCE: Ger. Offen., 16 pp.
 CODEN: GWXXBX
 DOCUMENT TYPE: Patent
 LANGUAGE: German
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
DE 10337830	A1	20050407	DE 2003-10337830	200308 18
DE 10337830	B4	20050825		200408 16
US 2005239272	A1	20051027	US 2004-920043	200308 18
PRIORITY APPLN. INFO.:			DE 2003-10337830	A

AB The fabrication of a multilayer structure, with metal layer on the surface of a 1st wafer and with a buffer layer on the metal layer, is described. A 2nd wafer is created on the buffer layer and the 1st wafer is then removed so that the metal layer is exposed.

IT 7440-21-3, Silicon, uses 11148-22-4
 RL: DEV (Device component use); USES (Uses)
 (fabrication of multilayer structure with exposed metal layer)

RN 7440-21-3 HCAPLUS

CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

RN 11148-22-4 HCAPLUS
 CN Germanium alloy, base, Ge,Si (9CI) (CA INDEX NAME)

Component	Component
Registry Number	
Ge	7440-56-4
Si	7440-21-3

IC ICM H01L021-283
 ICS H01L051-10; B81C001-00; G01N013-10
 CC 76-2 (Electric Phenomena)
 ST multilayer structure metal layer
 miniature switch
 IT Electric switches

(fabrication of multilayer structure with exposed metal layer)

IT Epoxy resins, uses
 RL: DEV (Device component use); USES (Uses)
 (fabrication of multilayer structure with exposed metal layer)

IT Electric contacts
 (multilayer; fabrication of multilayer structure with exposed metal layer)

IT 1303-00-0, Gallium arsenide (GaAs), uses 1312-41-0 1314-98-3, Zinc sulfide (ZnS), uses 1344-28-1, Aluminum oxide, uses 7440-05-3, Palladium, uses 7440-21-3, Silicon, uses 7440-32-6, Titanium, uses 7440-47-3, Chromium, uses 7440-58-6, Hafnium, uses 7631-86-9, Silicon oxide, uses 11148-22-4 12055-23-1, Hafnium oxide 12063-98-8, Gallium phosphide (GaP), uses 37382-15-3, Aluminum gallium arsenide 106070-25-1, Gallium indium arsenide 106311-99-3, Aluminum gallium phosphide
 RL: DEV (Device component use); USES (Uses)
 (fabrication of multilayer structure with exposed metal layer)

IT 12033-89-5, Silicon nitride, uses
 RL: TEM (Technical or engineered material use); USES (Uses)
 (fabrication of multilayer structure with exposed metal layer)

REFERENCE COUNT: 6 THERE ARE 6 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L29 ANSWER 4 OF 34 HCAPLUS COPYRIGHT 2006 ACS on STN
 ACCESSION NUMBER: 2004:789131 HCAPLUS
 DOCUMENT NUMBER: 142:229202
 TITLE: Strain relaxation of epitaxial CoSi₂ and SiGe layers in cap-Si/Si_{0.83}Ge_{0.17}/Si(001) and epi-CoSi₂/Si_{0.83}Ge_{0.17}/Si(001) structures
 AUTHOR(S): Shin, D. O.; Sardela, M. R., Jr.; Ban, S. H.; Lee, N.-E.; Shim, K.-H.
 CORPORATE SOURCE: Department of Materials Engineering, Center for Advanced Plasma Surface Technology, SungKyunKwan University, Kyunggi-do, 440-746, S. Korea
 SOURCE: Applied Surface Science (2004), 237(1-4), 139-145
 CODEN: ASUSEE; ISSN: 0169-4332
 PUBLISHER: Elsevier B.V.
 DOCUMENT TYPE: Journal
 LANGUAGE: English
 AB Strain relaxation behaviors of the epitaxial CoSi₂ (epi-CoSi₂) and Si_{0.83}Ge_{0.17} layers in epi-CoSi₂/Si_{0.83}Ge_{0.17}/Si(001) and cap-Si/Si_{0.83}Ge_{0.17}/Si(001) structures were investigated by high-resoln. X-ray diffraction (HR-XRD) analyses. Samples were treated at the temp., TA = 650-900° by rapid thermal annealing. Comparative measurements showed a different strain relaxation behavior in the SiGe layers with and without CoSi₂ layer. Ge content and lattice mismatch in the SiGe film of the epi-CoSi₂/Si_{0.83}Ge_{0.17}/Si(001) are smaller than those in the SiGe layer in cap-Si/SiGe/Si(001) possibly due to the diffusion of Ge into the tensile-stressed epi-CoSi₂ layer to reduce the compressive stress in the SiGe layer at elevated temp. The analyses of high-resoln. ω -2 θ scan spectra and reciprocal space mapping showed that epi-CoSi₂ layer is under tensile residual stress and a significant strain relaxation starts at TA = 900°

indicating of thermal stability up to TA = 850°.
 IT 7440-21-3, Silicon, properties 12017-12-8, Cobalt
 disilicide 113677-38-6, Germanium 17, silicon 83 (atomic)
 RL: PEP (Physical, engineering or chemical process); PRP
 (Properties); PYP (Physical process); PROC (Process)
 (strain relaxation of epitaxial CoSi₂ and SiGe layers in
 cap-Si/Si_{0.83}Ge_{0.17}/Si(001) and epi-CoSi₂/Si_{0.83}Ge_{0.17}/Si(001)
 structures)
 RN 7440-21-3 HCAPLUS
 CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

RN 12017-12-8 HCAPLUS
 CN Cobalt silicide (CoSi₂) (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)

Si
 |||
 Co≡Si

RN 113677-38-6 HCAPLUS
 CN Silicon alloy, base, Si 65,Ge 35 (9CI) (CA INDEX NAME)

Component	Component	Component
Percent	Registry Number	
Si	65	7440-21-3
Ge	35	7440-56-4

CC 75-1 (Crystallography and Liquid Crystals)
 ST cobalt silicide germanium silicon epitaxial multilayer
 strain relaxation
 IT Epitaxial films
 Mechanical relaxation
 Multilayers
 Rapid thermal annealing
 (strain relaxation of epitaxial CoSi₂ and SiGe layers in
 cap-Si/Si_{0.83}Ge_{0.17}/Si(001) and epi-CoSi₂/Si_{0.83}Ge_{0.17}/Si(001)
 structures)
 IT 7440-21-3, Silicon, properties 12017-12-8, Cobalt
 disilicide 113677-38-6, Germanium 17, silicon 83 (atomic)
 RL: PEP (Physical, engineering or chemical process); PRP
 (Properties); PYP (Physical process); PROC (Process)
 (strain relaxation of epitaxial CoSi₂ and SiGe layers in
 cap-Si/Si_{0.83}Ge_{0.17}/Si(001) and epi-CoSi₂/Si_{0.83}Ge_{0.17}/Si(001)
 structures)

REFERENCE COUNT: 11 THERE ARE 11 CITED REFERENCES AVAILABLE
 FOR THIS RECORD. ALL CITATIONS AVAILABLE
 IN THE RE FORMAT

L29 ANSWER 5 OF 34 HCAPLUS COPYRIGHT 2006 ACS on STN
 ACCESSION NUMBER: 2004:545809 HCAPLUS
 DOCUMENT NUMBER: 141:80575
 TITLE: Method for manufacture of strained semiconductor
 single crystals as channel layers for
 semiconductor multilayer structures as

metal oxide semiconductors for integrated circuits
 INVENTOR(S): Usami, Tokutaka; Ujihara, Toru; Fujiwara, Kozo; Nakajima, Kazuo
 PATENT ASSIGNEE(S): Tohoku University, Japan
 SOURCE: Jpn. Kokai Tokkyo Koho, 9 pp.
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 2004189505	A2	20040708	JP 2002-355674	200212 06
PRIORITY APPLN. INFO.:			JP 2002-355674	200212 06

AB The method includes prep. unstrained semiconductor single crystals A having lattice const. A', forming amorphous semiconductors B having lattice const. B' other than A', heating A and B for epitaxial growth of B, and solid-phase interdiffusion of A and B to give unstrained semiconductor mixed crystal C having lattice const. C', and formation of strained semiconductor single crystals D having lattice const. D' different from C' by epitaxial growth. Semiconductor **multilayer** structures include sequential layers of semiconductor single crystal substrates, insulators, unstrained semiconductor mixed crystals, and the strained semiconductor crystals. Strained semiconductor single crystals are manufd. more easily and economically than conventional methods using ion implantation app., thermal oxidn. app., etc.

IT 7440-21-3, Silicon, processes
 RL: CPS (Chemical process); DEV (Device component use); PEP (Physical, engineering or chemical process); TEM (Technical or engineered material use); PROC (Process); USES (Uses)
 (manuf. of strained semiconductor single crystals as channel layers for **multilayer** metal oxide semiconductors for integrated circuits)

RN 7440-21-3 HCPLUS
 CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

IT 11148-21-3P
 RL: DEV (Device component use); IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
 (manuf. of strained semiconductor single crystals as channel layers for **multilayer** metal oxide semiconductors for integrated circuits)

RN 11148-21-3 HCPLUS
 CN Germanium alloy, nonbase, Ge, Si (9CI) (CA INDEX NAME)

Component Component

Registry Number

=====

Ge	7440-56-4
Si	7440-21-3

IC ICM C30B001-10
 ICS C30B001-04; C30B029-52; H01L021-20; H01L021-203; H01L027-12
 CC 76-2 (Electric Phenomena)
 Section cross-reference(s): 75
 ST strained semiconductor single crystal MOS device; amorphous
 semiconductor unstrained single crystal interdiffusion;
 multilayer integrated circuit MOS strained semiconductor
 IT Semiconductor materials
 (layered; manuf. of strained semiconductor single crystals as
 channel layers for multilayer metal
 oxide semiconductors for integrated circuits)
 IT MOS devices
 (manuf. of strained semiconductor single crystals as channel
 layers for multilayer metal oxide
 semiconductors for integrated circuits)
 IT Integrated circuits
 (multilayer; manuf. of strained semiconductor single
 crystals as channel layers for multilayer
 metal oxide semiconductors for integrated circuits)
 IT 7440-56-4, Germanium, processes
 RL: CPS (Chemical process); PEP (Physical, engineering or chemical
 process); PROC (Process)
 (amorphous; manuf. of strained semiconductor single crystals as
 channel layers for multilayer metal
 oxide semiconductors for integrated circuits)
 IT 7440-21-3, Silicon, processes
 RL: CPS (Chemical process); DEV (Device component use); PEP
 (Physical, engineering or chemical process); TEM (Technical or
 engineered material use); PROC (Process); USES (Uses)
 (manuf. of strained semiconductor single crystals as channel
 layers for multilayer metal oxide
 semiconductors for integrated circuits)
 IT 11148-21-3P
 RL: DEV (Device component use); IMF (Industrial manufacture); TEM
 (Technical or engineered material use); PREP (Preparation); USES
 (Uses)
 (manuf. of strained semiconductor single crystals as channel
 layers for multilayer metal oxide
 semiconductors for integrated circuits)

L29 ANSWER 6 OF 34 HCPLUS COPYRIGHT 2006 ACS on STN

ACCESSION NUMBER: 2004:470757 HCPLUS
 DOCUMENT NUMBER: 141:32282
 TITLE: Formation of multilayer gate structure
 INVENTOR(S): Chen, Neng-Kuo; Akasaka, Yasushi
 PATENT ASSIGNEE(S): Hua-Pang Electronic Corp., Taiwan; Toshiba Corp.
 SOURCE: Jpn. Kokai Tokkyo Koho, 22 pp.
 CODEN: JKXXAF
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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JP 2004165174	A2	20040610	JP 2002-234990	200208
PRIORITY APPLN. INFO.:				12
JP 2002-234990				200208
				12

AB A method for forming a **multilayer** gate structure involves forming a Si layer doped with p-type ions, forming a shielding layer contg. a shielding component on the Si layer, and forming a **metal layer** on the shielding component via a bonding layer for decreasing the contact resistance of the gate structure while preventing the p-type ions from diffusing into the bonding layer with the shielding layer. Specifically, the p-type ions may comprise Br ions, the bonding layer may comprise W nitride, the **metal layer** may comprise W, and the shielding layer may contain Ge, Ge-Si, Mo, or Ta.

IT 7440-21-3, Silicon, uses 11148-21-3
 RL: DEV (Device component use); USES (Uses)
 (formation of **multilayer** gate structure having diffusion barrier)

RN 7440-21-3 HCPLUS

CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

RN 11148-21-3 HCPLUS
 CN Germanium alloy, nonbase, Ge,Si (9CI) (CA INDEX NAME)

Component	Component
Registry Number	
Ge	7440-56-4
Si	7440-21-3

IC ICM H01L029-78
 ICS H01L029-423; H01L029-49
 CC 76-3 (Electric Phenomena)
 ST **multilayer** gate structure diffusion barrier
 IT Diffusion barrier
 MOS devices
 Semiconductor device fabrication
 (formation of **multilayer** gate structure having diffusion barrier)
 IT 7439-98-7, Molybdenum, uses 7440-21-3, Silicon, uses 7440-25-7, Tantalum, uses 7440-33-7, Tungsten, uses 7440-56-4, Germanium, uses 11148-21-3 37359-53-8, Tungsten nitride
 RL: DEV (Device component use); USES (Uses)
 (formation of **multilayer** gate structure having diffusion barrier)
 IT 24959-67-9, Bromide, uses
 RL: MOA (Modifier or additive use); USES (Uses)
 (formation of **multilayer** gate structure having diffusion barrier)

L29 ANSWER 7 OF 34 HCPLUS COPYRIGHT 2006 ACS on STN

ACCESSION NUMBER: 2004:447296 HCAPLUS
 DOCUMENT NUMBER: 140:432656
 TITLE: Manufacture of semiconductor devices
 INVENTOR(S): Matsumura, Hiroaki
 PATENT ASSIGNEE(S): Sony Corp., Japan
 SOURCE: Jpn. Kokai Tokkyo Koho, 6 pp.
 CODEN: JKXXAF
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 2004158483	A2	20040603	JP 2002-319897	200211 01
PRIORITY APPLN. INFO.:				JP 2002-319897 200211 01

AB Semiconductor devices contain: substrates with buried diffusion layers; oxide films with openings on the substrates; epitaxial layers covering the oxides films and filling the openings; and silicide layers on the epitaxial layers. The silicide layers consist of 1st layers which are formed when metals absorbed Si during heat treatment, and 2nd layers which are formed on the 1st layers when the metals are diffused in Si during heat treatment.

IT 11148-21-3P
 RL: DEV (Device component use); PNU (Preparation, unclassified);
 PREP (Preparation); USES (Uses)
 (epitaxial layers; formation of metal
 silicide layers on epitaxial layers in manuf. of
 semiconductor devices)

RN 11148-21-3 HCAPLUS

CN Germanium alloy, nonbase, Ge,Si (9CI) (CA INDEX NAME)

Component	Component
Registry Number	

Ge	7440-56-4
Si	7440-21-3

IT 11104-62-4P, Cobalt silicide 12738-91-9P, Titanium silicide
 RL: DEV (Device component use); PNU (Preparation, unclassified);
 PREP (Preparation); USES (Uses)
 (formation of metal silicide layers in manuf.
 of semiconductor devices)

RN 11104-62-4 HCAPLUS

CN Cobalt silicide (9CI) (CA INDEX NAME)

Component	Ratio	Component
Registry Number		

Co	x	7440-48-4
Si	x	7440-21-3

RN 12738-91-9 HCAPLUS
 CN Titanium silicide (9CI) (CA INDEX NAME)

Component	Ratio	Component Registry Number
Ti	x	7440-32-6
Si	x	7440-21-3

IT 7440-21-3, Silicon, reactions
 RL: RCT (Reactant); RACT (Reactant or reagent)
 (formation of metal silicide layers in manuf.
 of semiconductor devices)

RN 7440-21-3 HCAPLUS
 CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

IC ICM H01L021-331
 ICS H01L021-28; H01L029-732
 CC 76-3 (Electric Phenomena)
 Section cross-reference(s): 75
 ST semiconductor device metal silicide multilayer; metal
 silicon absorption diffusion epitaxy
 IT Heat treatment
 (formation of metal silicide layers in manuf.
 of semiconductor devices)
 IT Silicides
 RL: DEV (Device component use); PNU (Preparation, unclassified);
 PREP (Preparation); USES (Uses)
 (formation of metal silicide layers in manuf.
 of semiconductor devices)
 IT Bipolar transistors
 Epitaxy
 (formation of metal silicide layers on
 epitaxial layers in manuf. of semiconductor devices)
 IT 11148-21-3P
 RL: DEV (Device component use); PNU (Preparation, unclassified);
 PREP (Preparation); USES (Uses)
 (epitaxial layers; formation of metal
 silicide layers on epitaxial layers in manuf. of
 semiconductor devices)
 IT 11104-62-4P, Cobalt silicide 12738-91-9P, Titanium
 silicide
 RL: DEV (Device component use); PNU (Preparation, unclassified);
 PREP (Preparation); USES (Uses)
 (formation of metal silicide layers in manuf.
 of semiconductor devices)
 IT 7440-21-3, Silicon, reactions
 RL: RCT (Reactant); RACT (Reactant or reagent)
 (formation of metal silicide layers in manuf.
 of semiconductor devices)

L29 ANSWER 8 OF 34 HCAPLUS COPYRIGHT 2006 ACS on STN
 ACCESSION NUMBER: 2004:414239 HCAPLUS
 DOCUMENT NUMBER: 141:131987
 TITLE: Strain modulation of β -FeSi₂ by
 Ge-segregation in solid-phase growth of

AUTHOR(S): [a-Si/a-FeSiGe]_n multi-layer
 Murakami, Y.; Kenjo, A.; Sadoh, T.; Yoshitake, T.; Itakura, M.; Miyao, M.

CORPORATE SOURCE: Department of Electronics, Kyushu University, Fukuoka, 812-8581, Japan

SOURCE: Materials Research Society Symposium Proceedings (2004), 796(Critical Interfacial Issues in Thin-Film Optoelectronic and Energy Conversion Devices), 57-62

CODEN: MRSPDH; ISSN: 0272-9172

PUBLISHER: Materials Research Society

DOCUMENT TYPE: Journal

LANGUAGE: English

AB Strain modulation of β -FeSi₂ by Ge doping was studied. By solid-phase growth of [a-Si/a-Fe_{0.4}Si_{0.5}Ge_{0.1}]_n layered structures, the [a-SiGe/ β -FeSi₂-xGex]_n multi-layered structures ($n = 1, 2, 4$) were formed after annealing at 700°. From the anal. of the x-ray diffraction spectra, β -FeSi_{1.3}Ge_{0.7} strained by 0.4-0.5% was formed for the sample with $n = 1$. This value corresponded to the band gap modulation of 30 meV based on the theor. calcn. The strains decreased with increasing n , which was due to that segregation of Ge atoms from the a-Fe_{0.4}Si_{0.5}Ge_{0.1} layers to the a-Si layers became significant with increasing n . After annealing at 800°, agglomeration of β -FeSi₂ occurred, and nanocrystals of relaxed β -FeSi₂ and c-Si_{0.7}Ge_{0.3} were formed. These new structures are useful for formation of opto-elec. devices.

IT 7440-21-3, Silicon, processes

RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PYP (Physical process); PROC (Process)
 ([a-SiGe/ β -FeSi₂-xGex]_n multi-layer
 formed by annealing of [a-Si/a-FeSiGe]_n multi-layer)

RN 7440-21-3 HCAPLUS

CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

IT 12623-04-0, Germanium 30 silicon 70 (atomic)

RL: FMU (Formation, unclassified); FORM (Formation, nonpreparative)
 (nanocrystals of relaxed β -FeSi₂ and c-Si_{0.7}Ge_{0.3} formed by annealing of [a-Si/a-FeSiGe]_n multi-layer)

RN 12623-04-0 HCAPLUS

CN Germanium alloy, base, Ge 53, Si 47 (9CI) (CA INDEX NAME)

Component	Component	Component
	Percent	Registry Number
Ge	53	7440-56-4
Si	47	7440-21-3

IT 12022-99-0, Iron silicide (FeSi₂)

RL: PRP (Properties); TEM (Technical or engineered material use); USES (Uses)

(strain modulation of β -FeSi₂ by Ge doping)

RN 12022-99-0 HCAPLUS

CN Iron silicide (FeSi₂) (6CI, 8CI, 9CI) (CA INDEX NAME)

Fe≡Si
 |||
 Si

CC 76-3 (Electric Phenomena)
 Section cross-reference(s): 75

IT Band gap
 (band gap modulation of β -FeSi₂ by Ge-segregation in solid-phase growth of [a-Si/a-FeSiGe]_n multi-layer)

IT Crystallization
 Optoelectronic semiconductor devices
 Strain
 (strain modulation of β -FeSi₂ by Ge-segregation in solid-phase growth of [a-Si/a-FeSiGe]_n multi-layer)

IT 7440-21-3, Silicon, processes
 RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PYP (Physical process); PROC (Process)
 ([a-SiGe/ β -FeSi₂-xGex]_n multi-layer
 formed by annealing of [a-Si/a-FeSiGe]_n multi-layer)

IT 721959-79-1, Germanium 7, iron 10, silicon 13 (atomic)
 RL: FMU (Formation, unclassified); PRP (Properties); FORM (Formation, nonpreparative)
 ([a-SiGe/ β -FeSi₂-xGex]_n multi-layer
 formed by annealing of [a-Si/a-FeSiGe]_n multi-layer)

IT 12623-04-0, Germanium 30 silicon 70 (atomic)
 RL: FMU (Formation, unclassified); FORM (Formation, nonpreparative)
 (nanocrystals of relaxed β -FeSi₂ and c-Si_{0.7}Ge_{0.3} formed by annealing of [a-Si/a-FeSiGe]_n multi-layer)

IT 12022-99-0, Iron silicide (FeSi₂)
 RL: PRP (Properties); TEM (Technical or engineered material use);
 USES (Uses)
 (strain modulation of β -FeSi₂ by Ge doping)

IT 642079-77-4, Germanium 10 iron 40 silicon 50 (atomic)
 RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PYP (Physical process); PROC (Process)
 (strain modulation of β -FeSi₂ by Ge-segregation in solid-phase growth of [a-Si/a-FeSiGe]_n multi-layer)

REFERENCE COUNT: 8 THERE ARE 8 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L29 ANSWER 9 OF 34 HCPLUS COPYRIGHT 2006 ACS on STN
 ACCESSION NUMBER: 2003:970570 HCPLUS
 DOCUMENT NUMBER: 140:279143
 TITLE: Self-aligned Ti germanosilicide formation on a polycrystalline Si/SiGe/Si extrinsic base for SiGe heterojunction bipolar transistors
 AUTHOR(S): Lee, Seung-yun; Park, Chan Woo; Kang, Jin-yeong
 CORPORATE SOURCE: Basic Research Laboratory, Electronics and Telecommunications Research Institute (ETRI), Daejeon, 305-350, S. Korea
 SOURCE: Journal of Electronic Materials (2003), 32(11),

1349-1356
 CODEN: JECMA5; ISSN: 0361-5235
 PUBLISHER: Minerals, Metals & Materials Society
 DOCUMENT TYPE: Journal
 LANGUAGE: English

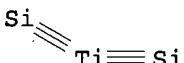
AB This work reports our investigation of a microstructure of self-aligned Ti germanosilicide made on polycryst. Si/SiGe/Si multilayers. The existence of the SiGe layer restricted the growth of the Ti germanosilicide layer and produced protrusions penetrating the underlying polycryst. layer. Each protrusion corresponded to a stacking-faulted single grain of the C49 phase. The microstructure of the thin Ti germanosilicide layer and the deep protrusions caused an increase of the sheet resistance and the contact resistivity of the extrinsic base region. The raised contact resistivity led to a degrdn. of radiofrequency (RF) and noise characteristics of the SiGe heterojunction bipolar transistor (HBT).
 IT 7440-21-3, Silicon, processes 11148-21-3
 RL: CPS (Chemical process); DEV (Device component use); PEP (Physical, engineering or chemical process); PROC (Process); USES (Uses)
 (self-aligned Ti germanosilicide formation on a polycryst. Si/SiGe/Si extrinsic base for SiGe heterojunction bipolar transistors)
 RN 7440-21-3 HCAPLUS
 CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

RN 11148-21-3 HCAPLUS
 CN Germanium alloy, nonbase, Ge,Si (9CI) (CA INDEX NAME)

Component	Component
Registry Number	
Ge	7440-56-4
Si	7440-21-3

IT 12039-83-7, Titanium disilicide
 RL: FMU (Formation, unclassified); FORM (Formation, nonpreparative)
 (self-aligned Ti germanosilicide formation on a polycryst. Si/SiGe/Si extrinsic base for SiGe heterojunction bipolar transistors)
 RN 12039-83-7 HCAPLUS
 CN Titanium silicide (TiSi₂) (6CI, 8CI, 9CI) (CA INDEX NAME)



CC 76-3 (Electric Phenomena)
 Section cross-reference(s): 75
 IT Contact resistance
 Electric noise
 Heterojunction bipolar transistors
 Microstructure
 Sheet resistance

Siliconizing
 Stacking faults
 (self-aligned Ti germanosilicide formation on a polycryst.
 Si/SiGe/Si extrinsic base for SiGe
 heterojunction bipolar transistors)
 IT 12355-90-7, Difluororoboron 1+
 RL: MOA (Modifier or additive use); USES (Uses)
 (ion implantation; self-aligned Ti germanosilicide formation on a
 polycryst. Si/SiGe/Si extrinsic base for SiGe
 heterojunction bipolar transistors)
 IT 7440-21-3, Silicon, processes 7440-32-6, Titanium,
 processes 7631-86-9, Silica, processes 11148-21-3
 25583-20-4, Titanium nitride TiN
 RL: CPS (Chemical process); DEV (Device component use); PEP
 (Physical, engineering or chemical process); PROC (Process); USES
 (Uses)
 (self-aligned Ti germanosilicide formation on a polycryst.
 Si/SiGe/Si extrinsic base for SiGe
 heterojunction bipolar transistors)
 IT 12039-83-7, Titanium disilicide 125135-18-4, Germanium
 titanium silicide ((Ge,Si)2Ti)
 RL: FMU (Formation, unclassified); FORM (Formation, nonpreparative)
 (self-aligned Ti germanosilicide formation on a polycryst.
 Si/SiGe/Si extrinsic base for SiGe
 heterojunction bipolar transistors)
 REFERENCE COUNT: 17 THERE ARE 17 CITED REFERENCES AVAILABLE
 FOR THIS RECORD. ALL CITATIONS AVAILABLE
 IN THE RE FORMAT

L29 ANSWER 10 OF 34 HCAPLUS COPYRIGHT 2006 ACS on STN
 ACCESSION NUMBER: 2003:553877 HCAPLUS
 DOCUMENT NUMBER: 140:208109
 TITLE: Epitaxial growth of CoSi₂ on Si(100)
 substrate by Co/GeSi/Ti/Si
 multilayer solid phase reaction
 AUTHOR(S): Xu, Beilei; Qu, Xinpeng; Han, Yongzhao; Ru,
 Guoping; Li, Bingzong; Cheung, W. Y.; Wong, S.
 P.; Chu, Paul K.
 CORPORATE SOURCE: Department of Microelectronics, ASIC and System
 State Key Laboratory, Fudan University,
 Shanghai, 200433, Peop. Rep. China
 SOURCE: Guti Dianzixue Yanjiu Yu Jinzhan (2003), 23(2),
 149-154
 CODEN: GDYJE2; ISSN: 1000-3819
 PUBLISHER: Guti Dianzixue Yanjiu Yu Jinzhan Bianjibu
 DOCUMENT TYPE: Journal
 LANGUAGE: Chinese
 AB The effect of amorphous Ge Si layer
 interposed between Co and Ti layers on Si (100)
 substrate on the solid phase heteroepitaxy of CoSi₂/Si
 structure was studied. Co/Ge-Si/Ti multilayers were
 sputtered on Si substrates. Epitaxial CoSi₂
 film with good elec. properties was formed after rapid thermal
 annealing. When the thickness of the interposed Ge-
 Si films changed from 0 to 10 nm, the formed CoSi₂
 film always has the favorable epitaxial quality and stable low elec.
 resistivity. At low annealing temp. (< 800°), Ti will
 combine with Co, O or Si to form ternary compds. such as Co₂Ti₄O and
 Ti₂Co₃Si, which will act as diffusion barrier and promote the
 epitaxial growth of CoSi₂ film. The Ge-Si interlayer will reduce

the consumption of substrate Si during CoSi₂ formation and the small amts. of Ge will improve the mismatch between epitaxial CoSi₂ and substrate Si.

IT 7440-21-3, Silicon, processes 37232-85-2,
Germanium 15, silicon 85 (atomic)
RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PYP (Physical process); TEM (Technical or engineered material use); PROC (Process); USES (Uses)
(epitaxy of cobalt silicide on silicon
substrate by cobalt/germanium-silicon/titanium/silicon
multilayer solid phase reaction)
RN 7440-21-3 HCAPLUS
CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

RN 37232-85-2 HCAPLUS
CN Silicon alloy, base, Si 69,Ge 31 (9CI) (CA INDEX NAME)

Component	Component Percent	Component Registry Number
Si	69	7440-21-3
Ge	31	7440-56-4

IT 12017-11-7, Cobalt silicide (CoSi)
RL: FMU (Formation, unclassified); FORM (Formation, nonpreparative)
(epitaxy of cobalt silicide on silicon
substrate by cobalt/germanium-silicon/titanium/silicon
multilayer solid phase reaction)
RN 12017-11-7 HCAPLUS
CN Cobalt silicide (CoSi) (6CI, 8CI, 9CI) (CA INDEX NAME)

Si
Co

IT 12017-12-8P, Cobalt silicide (CoSi₂)
RL: SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(epitaxy of cobalt silicide on silicon
substrate by cobalt/germanium-silicon/titanium/silicon
multilayer solid phase reaction)
RN 12017-12-8 HCAPLUS
CN Cobalt silicide (CoSi₂) (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)

Si
|||
Co=Si

CC 76-2 (Electric Phenomena)
Section cross-reference(s): 75
ST SPE cobalt silicide silicon substrate amorphous
germanium multilayer
IT Diffusion barrier

Multilayers
 Rapid thermal annealing
 Sheet resistance
 Siliconizing
 Solid phase epitaxy
 Thickness
 (epitaxy of cobalt silicide on silicon
 substrate by cobalt/germanium-silicon/titanium/silicon
 multilayer solid phase reaction)
IT Electric resistance
 (germanium-silicon; epitaxy of cobalt silicide on silicon
 substrate by cobalt/germanium-silicon/titanium/silicon
 multilayer solid phase reaction)
IT 7440-21-3, Silicon, processes 7440-32-6, Titanium,
 processes 7440-48-4, Cobalt, processes 37232-85-2,
 Germanium 15, silicon 85 (atomic)
 RL: CPS (Chemical process); PEP (Physical, engineering or chemical
 process); PYP (Physical process); TEM (Technical or engineered
 material use); PROC (Process); USES (Uses)
 (epitaxy of cobalt silicide on silicon
 substrate by cobalt/germanium-silicon/titanium/silicon
 multilayer solid phase reaction)
IT 12017-11-7, Cobalt silicide (CoSi) 12052-56-1, Cobalt
 titanium silicide (Co₃Ti₂Si) 61179-90-6, Cobalt titanium oxide
 (Co₂Ti₄O)
 RL: FMU (Formation, unclassified); FORM (Formation, nonpreparative)
 (epitaxy of cobalt silicide on silicon
 substrate by cobalt/germanium-silicon/titanium/silicon
 multilayer solid phase reaction)
IT 12017-12-8P, Cobalt silicide (CoSi₂)
 RL: SPN (Synthetic preparation); TEM (Technical or engineered
 material use); PREP (Preparation); USES (Uses)
 (epitaxy of cobalt silicide on silicon
 substrate by cobalt/germanium-silicon/titanium/silicon
 multilayer solid phase reaction)

L29 ANSWER 11 OF 34 HCPLUS COPYRIGHT 2006 ACS on STN
ACCESSION NUMBER: 2003:50434 HCPLUS
DOCUMENT NUMBER: 138:359668
TITLE: Femtosecond pump-probe nondestructive
 examination of materials (invited)
AUTHOR(S): Norris, Pamela M.; Caffrey, Andrew P.; Stevens,
 Robert J.; Klopff, J. Michael; McLeskey, James
 T., Jr.; Smith, Andrew N.
CORPORATE SOURCE: Department of Mechanical and Aerospace
 Engineering, University of Virginia,
 Charlottesville, VA, 22904, USA
SOURCE: Review of Scientific Instruments (2003), 74(1,
 Pt. 2), 400-406
 CODEN: RSINAK; ISSN: 0034-6748
PUBLISHER: American Institute of Physics
DOCUMENT TYPE: Journal
LANGUAGE: English
AB Ultrashort-pulsed lasers were demonstrated as effective tools for
 the nondestructive examn. (NDE) of energy transport properties in
 thin films. After the instantaneous heating of the surface of a
 100. nm metal film, it will take .apprx.100 ps
 for the influence of the substrate to affect the surface temp.
 profile. Therefore, direct measurement of energy transport in a
 thin film sample requires a technique with picosecond temporal

resoln. The pump-probe exptl. technique is able to monitor the change in reflectance or transmittance of the sample surface as a function of time on a subpicosecond time scale. Changes in reflectance and transmittance can then be used to det. properties of the film. In the case of metals, the change in reflectance is related to changes in temp. and strain. The transient temp. profile at the surface is then used to det. the rate of coupling between the electron and phonon systems as well as the thermal cond. of the material. In the case of semiconductors, the change in reflectance and transmittance is related to changes in the local electronic states and temp. Transient thermotransmission expts. were used extensively to observe electron-hole recombination phenomena and thermalization of hot electrons. Application of the transient thermoreflectance (TTR) and transient thermotransmittance (TTT) technique to the study of picosecond phenomena in metals and semiconductors are discussed. The pump-probe exptl. setup will be described, along with the details of the exptl. app. in use at the University of Virginia. The thermal model applicable to ultrashort-pulsed laser heating of metals will be presented along with a discussion of the limitations of this model. Details of the data acquisition and interpretation of the exptl. results will be given, including a discussion of the reflectance models used to relate the measured changes in reflectance to calcd. changes in temp. Finally, exptl. results will be presented that demonstrate the use of the TTR technique for measuring the electron-phonon coupling factor and the thermal cond. of thin metallic films. The use of the TTT technique to distinguish between different levels of doping and alloying in thin film samples of hydrogenated amorphous Si will also be discussed briefly.

IT 7440-21-3, Silicon, properties 11148-21-3
 RL: PRP (Properties)
 (hydrogenated; femtosecond pump-probe nondestructive examn. of)

RN 7440-21-3 HCAPLUS

CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

RN 11148-21-3 HCAPLUS
 CN Germanium alloy, nonbase, Ge,Si (9CI) (CA INDEX NAME)

Component	Component
Registry Number	
Ge	7440-56-4
Si	7440-21-3

CC 73-2 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)

Section cross-reference(s): 69

ST pump optical probe material thermal cond energy transfer; platinum multilayer gold optical pump probe conductive energy transfer; hydrogenated germanium silicon optical pump probe conductive energy transfer

IT Multilayers
 (femtosecond pump-probe nondestructive examn. of materials with metal)

IT 7440-21-3, Silicon, properties 11148-21-3
 RL: PRP (Properties)

(hydrogenated; femtosecond pump-probe nondestructive examn. of)
 REFERENCE COUNT: 45 THERE ARE 45 CITED REFERENCES AVAILABLE
 FOR THIS RECORD. ALL CITATIONS AVAILABLE
 IN THE RE FORMAT

L29 ANSWER 12 OF 34 HCAPLUS COPYRIGHT 2006 ACS on STN
 ACCESSION NUMBER: 2002:814606 HCAPLUS
 DOCUMENT NUMBER: 137:319080
 TITLE: MOSFET integrated circuit with thin film having
 high permittivity and uniform thickness
 INVENTOR(S): Yamamoto, Ichiro
 PATENT ASSIGNEE(S): NEC Corp., Japan
 SOURCE: U.S. Pat. Appl. Publ., 20 pp.
 CODEN: USXXCO
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 2002153579	A1	20021024	US 2002-125370	200204 19
JP 2002314072	A2	20021025	JP 2001-120485	200104 19
PRIORITY APPLN. INFO.:			JP 2001-120485	A 200104 19

AB The invention relates to a MOSFET integrated circuit with a thin film having a high dielec. const. and uniform film thickness. The semiconductor device comprises, in an embodiment, an electrode which is made of a metal or a metal nitride and which is formed on a silicon layer via a dielec. film. The dielec. film has a multi-layer structure comprising an amorphous oxide film on the side of the silicon layer and a metal oxide film on the side of the electrode. In another embodiment, the semiconductor device comprises an electrode which is made of silicon (Si) or a silicon germanium (SiGe) and which is formed on a silicon layer via a dielec. film. In such case, the dielec. film has a multilayer structure comprising a first amorphous oxide film on the side of the silicon layer, a second amorphous oxide film on the side of the electrode, and a metal oxide film between the first and second amorphous oxide films.

IT 7440-21-3, Silicon, processes 11148-21-3
 RL: DEV (Device component use); EPR (Engineering process); PEP (Physical, engineering or chemical process); PROC (Process); USES (Uses)

(MOSFET integrated circuit with thin film having high permittivity and uniform thickness)

RN 7440-21-3 HCAPLUS

CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

RN 11148-21-3 HCPLUS
CN Germanium alloy, nonbase, Ge,Si (9CI) (CA INDEX NAME)

Component	Component Registry Number
Ge	7440-56-4
Si	7440-21-3

IC ICM H01L029-76
IC S H01L021-336; H01L029-94; H01L031-062; H01L031-113; H01L021-3205
INCL 257412000
CC 76-3 (Electric Phenomena)
IT 1314-23-4, Zirconia, processes 1344-28-1, Alumina, processes
7440-21-3, Silicon, processes 7631-86-9, Silica, processes
11148-21-3
RL: DEV (Device component use); EPR (Engineering process); PEP
(Physical, engineering or chemical process); PROC (Process); USES
(Uses)
 (MOSFET integrated circuit with thin film having high
 permittivity and uniform thickness)

L29 ANSWER 13 OF 34 HCAPLUS COPYRIGHT 2006 ACS on STN

ACCESSION NUMBER: 2001:690110 HCAPLUS

ACCESSION NUMBER: 2001-09011
DOCUMENT NUMBER: 135:234772

TITLE: Forming a conductive multilayer

TITLE: Forming a conductive multilayer structure in a semiconductor device
INVENTOR(S): Weimer, Ronald A.; Hu, Yongjun Jeff; Pan, Pai Hung; Ratakonda, Deepa; Beck, James; Thakur, Pandhir P. S.

PATENT ASSIGNEE(S): Randhawa P. S. Micron Technology, Inc. USA

PATENT ASSIGNEE (S) : MICRON Techni
SOURCE: U.S. 12 pp

SOURCE: U.S., 12 pp.
DOCUMENT TYPE: CODEN: USXXAM
Patent

DOCUMENT TYPE: patent
LANGUAGE: English

LANGUAGE: English
FAMILY ACC NUM COUNT: 3

FAMILY ACC. NUM. COUNT: 2
PATENT INFORMATION

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 6291868	B1	20010918	US 1998-31407	19980226
US 2001014522	A1	20010816	US 1999-397763	19990915
US 6362086	B2	20020326		
US 6596595	B1	20030722	US 2000-620442	20000720
US 2003207556	A1	20031106	US 2003-454218	20030604
US 6849544	B2	20050201	US 1998-31407	A3 19980226
PRIORITY APPLN. INFO.:				

US 1999-397763 A1
199909
15

US 2000-620442 A1
200007
20

AB A conductive structure for use in a semiconductor device includes a multilayer structure. A 1st layer includes a material contg. Si, e.g., polysilicon and Si-Ge. A barrier layer is formed over the 1st layer, with the barrier layer including metal silicide or metal silicide nitride. A top conductive layer is formed over the barrier layer. The top conductive layer can include metal or metal silicide. Selective oxidn. can be performed to reduce the amt. of oxidn. of selected materials in a structure contg. multiple layers, such as the multilayer conductive structure. The selective oxidn. was performed in a single-wafer rapid thermal processing system, in which a selected ambient, including H, was used to ensure low oxidn. of a selected material, such as W or a metal nitride.

IT 7440-21-3, Silicon, processes 11104-62-4, Cobalt silicide 11148-21-3 12035-57-3, Nickel silicide (NiSi) 12738-91-9, Titanium silicide 39467-10-2, Nickel silicide

RL: DEV (Device component use); PEP (Physical, engineering or chemical process); PROC (Process); USES (Uses)
(forming conductive multilayer structure in semiconductor device)

RN 7440-21-3 HCAPLUS

CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

RN 11104-62-4 HCAPLUS

CN Cobalt silicide (9CI) (CA INDEX NAME)

Component	Ratio	Component Registry Number
Co	x	7440-48-4
Si	x	7440-21-3

RN 11148-21-3 HCAPLUS

CN Germanium alloy, nonbase, Ge,Si (9CI) (CA INDEX NAME)

Component	Component Registry Number
Ge	7440-56-4
Si	7440-21-3

RN 12035-57-3 HCAPLUS

CN Nickel silicide (NiSi) (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)

Ni \equiv Si

RN 12738-91-9 HCAPLUS
 CN Titanium silicide (9CI) (CA INDEX NAME)

Component	Ratio	Component Registry Number
Ti	x	7440-32-6
Si	x	7440-21-3

RN 39467-10-2 HCAPLUS
 CN Nickel silicide (9CI) (CA INDEX NAME)

Component	Ratio	Component Registry Number
Si	x	7440-21-3
Ni	x	7440-02-0

IC ICM H01L029-76
 INCL 257413000
 CC 76-2 (Electric Phenomena)
 ST conductor multilayer silicide nitride
 IT Films
 (elec. conductive; forming conductive multilayer structure in semiconductor device)
 IT Electric conductors
 (films; forming conductive multilayer structure in semiconductor device)
 IT Diffusion barrier
 Electric conductors
 Multilayers
 (forming conductive multilayer structure in semiconductor device)
 IT Metals, processes
 Transition metal nitrides
 Transition metal silicides
 RL: DEV (Device component use); PEP (Physical, engineering or chemical process); PROC (Process); USES (Uses)
 (forming conductive multilayer structure in semiconductor device)
 IT Transition metal nitrides
 Transition metal silicides
 RL: DEV (Device component use); PEP (Physical, engineering or chemical process); PROC (Process); USES (Uses)
 (nitride silicides; forming conductive multilayer structure in semiconductor device)
 IT Oxidation
 (selective; forming conductive multilayer structure in semiconductor device)
 IT 7440-21-3, Silicon, processes 7440-33-7, Tungsten, processes 11104-62-4, Cobalt silicide 11104-85-1, Molybdenum silicide 11148-21-3 12035-57-3, Nickel silicide (NiSi) 12738-91-9, Titanium silicide 39336-13-5, Niobium silicide 39467-10-2, Nickel silicide 60304-33-8, Hafnium silicide
 RL: DEV (Device component use); PEP (Physical, engineering or

chemical process); PROC (Process); USES (Uses)
 (forming conductive multilayer structure in
 semiconductor device)

IT 1333-74-0, Hydrogen, uses
 RL: NUU (Other use, unclassified); USES (Uses)
 (forming conductive multilayer structure in
 semiconductor device)

REFERENCE COUNT: 16 THERE ARE 16 CITED REFERENCES AVAILABLE
 FOR THIS RECORD. ALL CITATIONS AVAILABLE
 IN THE RE FORMAT

L29 ANSWER 14 OF 34 HCAPLUS COPYRIGHT 2006 ACS on STN
 ACCESSION NUMBER: 2001:598452 HCAPLUS
 DOCUMENT NUMBER: 135:160980
 TITLE: Forming a conductive structure
 multilayer including metal silicides in
 a semiconductor device
 INVENTOR(S): Weimer, Ronald A.; Hu, Yongjun Jeff; Pan, Pai
 Hung; Ratakonda, Deepa; Beck, James; Thakur,
 Randhir P. S.
 PATENT ASSIGNEE(S): Micron Technology, Inc., USA
 SOURCE: U.S. Pat. Appl. Publ., 12 pp., Division of U.S.
 Ser. No. 31,407.
 CODEN: USXXCO
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 2
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 2001014522	A1	20010816	US 1999-397763	199909 15
US 6362086	B2	20020326		
US 6291868	B1	20010918	US 1998-31407	199802 26
US 6596595	B1	20030722	US 2000-620442	200007 20
US 2003207556	A1	20031106	US 2003-454218	200306 04
US 6849544	B2	20050201	US 1998-31407	A3 199802 26
PRIORITY APPLN. INFO.:			US 1999-397763	A1 199909 15
			US 2000-620442	A1 200007 20

AB A conductive structure for use in a semiconductor device includes a
 multilayer structure. A 1st layer includes a material

contg. Si, e.g., polysilicon and Si germanide. A barrier layer is formed over the 1st layer, with the barrier layer including metal silicide or metal silicide nitride. A top conductive layer is formed over the barrier layer. The top conductive layer can include metal or metal silicide. Selective oxidn. can be performed to reduce the amt. of oxidn. of selected materials in a structure contg. **multiple layers**, such as the **multilayer** conductive structure. The selective oxidn. was performed in a single-wafer rapid thermal processing system, in which a selected ambient, including H, was used to ensure low oxidn. of a selected material, such as W or a metal nitride.

IT 7440-21-3, Silicon, processes 11148-21-3
 RL: PEP (Physical, engineering or chemical process); TEM (Technical or engineered material use); PROC (Process); USES (Uses)
 (forming conductive structure **multilayer** including metal silicides in semiconductor device)

RN 7440-21-3 HCAPLUS
 CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

RN 11148-21-3 HCAPLUS
 CN Germanium alloy, nonbase, Ge,Si (9CI) (CA INDEX NAME)

Component	Component Registry Number
Ge	7440-56-4
Si	7440-21-3

IT 11104-62-4P, Cobalt silicide 11113-78-3P, Palladium silicide 12738-91-9P, Titanium silicide 39467-10-2P, Nickel silicide
 RL: PNU (Preparation, unclassified); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
 (forming conductive structure **multilayer** including metal silicides in semiconductor device)

RN 11104-62-4 HCAPLUS
 CN Cobalt silicide (9CI) (CA INDEX NAME)

Component	Ratio	Component Registry Number
Co	x	7440-48-4
Si	x	7440-21-3

RN 11113-78-3 HCAPLUS
 CN Palladium silicide (9CI) (CA INDEX NAME)

Component	Ratio	Component Registry Number
Si	x	7440-21-3
Pd	x	7440-05-3

RN 12738-91-9 HCAPLUS
 CN Titanium silicide (9CI) (CA INDEX NAME)

Component	Ratio	Component Registry Number
Ti	x	7440-32-6
Si	x	7440-21-3

RN 39467-10-2 HCAPLUS
 CN Nickel silicide (9CI) (CA INDEX NAME)

Component	Ratio	Component Registry Number
Si	x	7440-21-3
Ni	x	7440-02-0

IC ICM H01L021-3205
 ICS H01L021-4763; H01L021-44
 INCL 438591000
 CC 76-2 (Electric Phenomena)
 ST conductor **multilayer** silicide nitride semiconductor device
 IT Vapor deposition process
 (chem.; forming conductive structure **multilayer**
 including metal silicides in semiconductor device)
 IT Films
 (elec. conductive; forming conductive structure
 multilayer including metal silicides in semiconductor
 device)
 IT Electric conductors
 (films; forming conductive structure **multilayer**
 including metal silicides in semiconductor device)
 IT Diffusion barrier
 Electric conductors
 Semiconductor devices
 (forming conductive structure **multilayer** including
 metal silicides in semiconductor device)
 IT Transition metals, processes
 RL: NUU (Other use, unclassified); PEP (Physical, engineering or
 chemical process); TEM (Technical or engineered material use); PROC
 (Process); USES (Uses)
 (forming conductive structure **multilayer** including
 metal silicides in semiconductor device)
 IT Transition metal silicides
 RL: PEP (Physical, engineering or chemical process); PNU
 (Preparation, unclassified); TEM (Technical or engineered material
 use); PREP (Preparation); PROC (Process); USES (Uses)
 (forming conductive structure **multilayer** including
 metal silicides in semiconductor device)
 IT Transition metal nitrides
 RL: PEP (Physical, engineering or chemical process); TEM (Technical
 or engineered material use); PROC (Process); USES (Uses)
 (forming conductive structure **multilayer** including
 metal silicides in semiconductor device)
 IT Transition metal nitrides
 Transition metal silicides
 RL: PEP (Physical, engineering or chemical process); PNU
 (Preparation, unclassified); TEM (Technical or engineered material
 use); PREP (Preparation); PROC (Process); USES (Uses)
 (nitride silicides; forming conductive structure
 multilayer including metal silicides in semiconductor

device)
 IT Vapor deposition process
 (phys.; forming conductive structure **multilayer**
 including metal silicides in semiconductor device)
 IT Oxidation
 (selective; forming conductive structure **multilayer**
 including metal silicides in semiconductor device)
 IT 1333-74-0, Hydrogen, uses 7732-18-5, Water, uses 7782-44-7,
 Oxygen, uses
 RL: NUU (Other use, unclassified); USES (Uses)
 (forming conductive structure **multilayer** including
 metal silicides in semiconductor device)
 IT 302-01-2, Hydrazine, processes 7664-41-7, Ammonia, processes
 7727-37-9, Nitrogen, processes
 RL: NUU (Other use, unclassified); PEP (Physical, engineering or
 chemical process); PROC (Process); USES (Uses)
 (forming conductive structure **multilayer** including
 metal silicides in semiconductor device)
 IT 7440-21-3, Silicon, processes 7440-33-7, Tungsten,
 processes 11148-21-3
 RL: PEP (Physical, engineering or chemical process); TEM (Technical
 or engineered material use); PROC (Process); USES (Uses)
 (forming conductive structure **multilayer** including
 metal silicides in semiconductor device)
 IT 11104-62-4P, Cobalt silicide 11113-78-3P,
 Palladium silicide 12738-91-9P, Titanium silicide
 39467-10-2P, Nickel silicide
 RL: PNU (Preparation, unclassified); TEM (Technical or engineered
 material use); PREP (Preparation); USES (Uses)
 (forming conductive structure **multilayer** including
 metal silicides in semiconductor device)

L29 ANSWER 15 OF 34 HCAPLUS COPYRIGHT 2006 ACS on STN
 ACCESSION NUMBER: 2001:400628 HCAPLUS
 DOCUMENT NUMBER: 135:160574
 TITLE: Three-dimensional nano-objects evolving from a
 two-dimensional layer technology
 AUTHOR(S): Schmidt, Oliver G.; Schmarje, Nicole; Deneke,
 Christoph; Muller, Claudia; Jin-Phillipp,
 Neng-Yun
 CORPORATE SOURCE: Max-Planck-Institut fur Festkorperforschung,
 Stuttgart, D-70569, Germany
 SOURCE: Advanced Materials (Weinheim, Germany) (2001),
 13(10), 756-759
 CODEN: ADVMEW; ISSN: 0935-9648
 PUBLISHER: Wiley-VCH Verlag GmbH
 DOCUMENT TYPE: Journal
 LANGUAGE: English
 AB Nanotubes were formed from a thin solid film which is peeled off
 from the substrate by a selective etching procedure. The rolling up
 of the film is caused by an inherently built-in strain. In this
 way, self-erecting, vertical, ultrathin SiGe membranes were
 fabricated. A **multilayer** structure is deposited on a
 Si(001) substrate by MBE. It consists of a Ge
 sacrificial buffer layer and 2 SiGe layers with different amts. of
 Ge. The etchant soln. (H₂O₂/H₂O) is introduced through a slit on
 the sample surface. The microstructure of the samples was
 investigated by SEM. The built-in strain in the SiGe bilayer was
 further utilized by sepg. the free-standing membranes from their
 base. In this way, a ring-like, vertical membrane is formed on the

substrate surface. The nanofabrication process was also used for the prodn. of a multi-wall InGaAs nanotube rolled up from a GaAs(001) substrate. Furthermore, it was possible to combine different materials in a single nanotube, which was demonstrated for a nanotube the walls of which contained semiconductor (SiGe), insulator (SiO₂), and metal (Ti) layers.

IT 11148-21-3

RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process)

(prodn. of SiGe nanotubes evolving from a 2D layer technol.)

RN 11148-21-3 HCPLUS

CN Germanium alloy, nonbase, Ge,Si (9CI) (CA INDEX NAME)

Component	Component
Registry Number	
Ge	7440-56-4
Si	7440-21-3

IT 37380-03-3, germanium 20, silicon 80 (atomic)
 76998-02-2, germanium 40, silicon 60 (atomic)
 83573-93-7, germanium 70, silicon 30 (atomic)

RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process)

(prodn. of SiGe/SiO_x/Ti nanotubes evolving from a 2D layer technol. on Ge buffer)

RN 37380-03-3 HCPLUS

CN Silicon alloy, base, Si 61,Ge 39 (9CI) (CA INDEX NAME)

Component	Component	Component
Percent	Registry Number	
Si	61	7440-21-3
Ge	39	7440-56-4

RN 76998-02-2 HCPLUS

CN Germanium alloy, base, Ge 63,Si 37 (9CI) (CA INDEX NAME)

Component	Component	Component
Percent	Registry Number	
Ge	63	7440-56-4
Si	37	7440-21-3

RN 83573-93-7 HCPLUS

CN Germanium alloy, base, Ge 86,Si 14 (9CI) (CA INDEX NAME)

Component	Component	Component
Percent	Registry Number	
Ge	86	7440-56-4
Si	14	7440-21-3

IT 7440-21-3, Silicon, uses

RL: NUU (Other use, unclassified); USES (Uses)
 (substrate; prodn. of SiGe nanotubes evolving from a 2D layer technol.)

RN 7440-21-3 HCPLUS

CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

CC 76-2 (Electric Phenomena)
 IT 11148-21-3
 RL: PEP (Physical, engineering or chemical process); PRP
 (Properties); PROC (Process)
 (prodn. of SiGe nanotubes evolving from a 2D layer technol.)
 IT 7440-32-6, Titanium, properties 7440-56-4, Germanium, properties
 7631-86-9, Silica, properties 37380-03-3, germanium 20,
 silicon 80 (atomic) 76998-02-2, germanium 40, silicon 60
 (atomic) 83573-93-7, germanium 70, silicon 30 (atomic)
 RL: PEP (Physical, engineering or chemical process); PRP
 (Properties); PROC (Process)
 (prodn. of SiGe/SiO_x/Ti nanotubes evolving from a 2D layer
 technol. on Ge buffer)
 IT 7440-21-3, Silicon, uses
 RL: NUU (Other use, unclassified); USES (Uses)
 (substrate; prodn. of SiGe nanotubes evolving from a 2D
 layer technol.)
 REFERENCE COUNT: 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR
 THIS RECORD. ALL CITATIONS AVAILABLE IN
 THE RE FORMAT

L29 ANSWER 16 OF 34 HCAPLUS COPYRIGHT 2006 ACS on STN
 ACCESSION NUMBER: 2001:150628 HCAPLUS
 DOCUMENT NUMBER: 134:200314
 TITLE: Efficient silicon-germanium infrared detectors
 INVENTOR(S): Presting, Hartmut; Jaros, Milan
 PATENT ASSIGNEE(S): Daimlerchrysler A.-G., Germany
 SOURCE: Ger. Offen., 6 pp.
 CODEN: GWXXBX
 DOCUMENT TYPE: Patent
 LANGUAGE: German
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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DE 19941304	A1	20010301	DE 1999-19941304	199908 31

PRIORITY APPLN. INFO.:	DE 1999-19941304	199908 31
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AB Multilayer structures for IR detectors are described which
 comprise a Si substrate, a lightly doped Si
 layer, an intrinsic first Si_{1-x}Ge_x epitaxial layer, a heavily doped
 second Si_{1-x}Ge_x epitaxial layer, another intrinsic third Si_{1-x}Ge_x
 epitaxial layer, and a metal (e.g., Al) contact.
 The third Si_{1-x}Ge_x epitaxial layer acts as a dopant setback layer ,
 producing a gap in the charge carrier flow from the metal to the
 semiconductor ad resulting in ballistic carrier injection. This
 reduces the dark current and increases the efficiency of the
 detector.

IT 7440-21-3, Silicon, uses 12623-02-8, Germanium 50,
 silicon 50 (atomic) 12623-04-0, Germanium 30, silicon 70

(atomic)
 RL: DEV (Device component use); USES (Uses)
 (silicon-germanium IR detectors with ballistic carrier injection)
 RN 7440-21-3 HCPLUS
 CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

RN 12623-02-8 HCPLUS
 CN Germanium alloy, base, Ge 72, Si 28 (9CI) (CA INDEX NAME)

Component	Component Percent	Component Registry Number
Ge	72	7440-56-4
Si	28	7440-21-3

RN 12623-04-0 HCPLUS
 CN Germanium alloy, base, Ge 53, Si 47 (9CI) (CA INDEX NAME)

Component	Component Percent	Component Registry Number
Ge	53	7440-56-4
Si	47	7440-21-3

IC ICM H01L031-109
 ICS G01J005-20

CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)

Section cross-reference(s): 76

IT 7429-90-5, Aluminum, uses 7440-21-3, Silicon, uses
 12623-02-8, Germanium 50, silicon 50 (atomic)
 12623-04-0, Germanium 30, silicon 70 (atomic)

RL: DEV (Device component use); USES (Uses)
 (silicon-germanium IR detectors with ballistic carrier injection)

REFERENCE COUNT: 7 THERE ARE 7 CITED REFERENCES AVAILABLE FOR
 THIS RECORD. ALL CITATIONS AVAILABLE IN
 THE RE FORMAT

L29 ANSWER 17 OF 34 HCPLUS COPYRIGHT 2006 ACS on STN

ACCESSION NUMBER: 2000:881466 HCPLUS

DOCUMENT NUMBER: 134:35989

TITLE: Non-volatile semiconductor memory cell,
 comprising a metal-oxide dielectric, and a
 method for producing the same.

INVENTOR(S): Ludwig, Christoph; Schrems, Martin

PATENT ASSIGNEE(S): Infineon Technologies A.-G., Germany

SOURCE: PCT Int. Appl., 29 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent

LANGUAGE: German

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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WO 2000075997	A1	20001214	WO 2000-DE1866	
				200006
				06
W: JP, KR, US				
RW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE				
DE 19926108	A1	20001221	DE 1999-19926108	
				199906
				08
EP 1183735	A1	20020306	EP 2000-947793	
				200006
				06
EP 1183735	B1	20051019		
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, FI				
US 2002093858	A1	20020718	US 2001-13271	
				200112
				10
US 6580118	B2	20030617		
PRIORITY APPLN. INFO.:			DE 1999-19926108	A
				199906
				08
			WO 2000-DE1866	W
				200006
				06

AB The invention relates to a nonvolatile semiconductor memory cell and a method for producing the same. In the method, a conventional, dielec. ONO layer is replaced by an extremely thin metal-oxide layer, consisting of WO_x and/or TiO₂. An addnl. improvement in the integration d. and the control voltage necessary for the semiconductor memory cell is achieved as a result of the high relative dielec. const. of these materials.

IT 7440-21-3, Silicon, processes 11148-21-3
 RL: DEV (Device component use); PEP (Physical, engineering or chemical process); TEM (Technical or engineered material use); PROC (Process); USES (Uses)
 (non-volatile semiconductor memory cell, comprising a metal-oxide dielec., and a method for producing the same.)

RN 7440-21-3 HCPLUS

CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

RN 11148-21-3 HCPLUS
 CN Germanium alloy, nonbase, Ge, Si (9CI) (CA INDEX NAME)

Component	Component
Registry Number	
Ge	7440-56-4
Si	7440-21-3

IC ICM H01L029-51
 ICS H01L029-788

CC 76-3 (Electric Phenomena)

ST nonvolatile semiconductor memory cell metal oxide dielec prodn;

metal oxide layer improvement integration density
 control voltage EEPROM
 IT Dielectric constant
 Electric insulators
 Electric potential
 Etching
 Multilayers
 SOI devices
 Semiconductor memory devices
 Sputtering
 Volatility
 (non-volatile semiconductor memory cell, comprising a metal-oxide dielec., and a method for producing the same.)
 IT 7440-32-6, Titanium, processes
 RL: DEV (Device component use); PEP (Physical, engineering or chemical process); TEM (Technical or engineered material use); PROC (Process); USES (Uses)
 (metal layer; non-volatile semiconductor memory cell, comprising a metal-oxide dielec., and a method for producing the same.)
 IT 78-10-4, TEOS 409-21-2, Silicon carbide, processes 1303-00-0,
 Gallium arsenide, processes 7440-21-3, Silicon, processes
 7440-33-7, Tungsten, processes 7631-86-9, Silica, processes
 11148-21-3 12033-89-5, Silicon nitride (Si₃N₄), processes
 12627-41-7, Tungsten silicide 13463-67-7, Titania, processes
 25583-20-4, Titanium nitride 37359-53-8, Tungsten nitride
 RL: DEV (Device component use); PEP (Physical, engineering or chemical process); TEM (Technical or engineered material use); PROC (Process); USES (Uses)
 (non-volatile semiconductor memory cell, comprising a metal-oxide dielec., and a method for producing the same.)

REFERENCE COUNT: 6 THERE ARE 6 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L29 ANSWER 18 OF 34 HCAPLUS COPYRIGHT 2006 ACS on STN
 ACCESSION NUMBER: 2000:877063 HCAPLUS
 DOCUMENT NUMBER: 134:65332
 TITLE: Perpendicular magnetic recording medium and magnetic recording/reproduction apparatus
 INVENTOR(S): Nimoto, Masaaki; Honda, Yukio; Hirayama, Yoshiyuki; Kikukawa, Atsushi; Yoshida, Kazuyoshi
 PATENT ASSIGNEE(S): Hitachi, Ltd., Japan
 SOURCE: Jpn. Kokai Tokkyo Koho, 9 pp.
 CODEN: JKXXAF
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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JP 2000348327	A2	20001215	JP 1999-162024	199906 09
PRIORITY APPLN. INFO.:			JP 1999-162024	199906 09

AB In a perpendicular magnetic recording medium having a perpendicular magnetization film formed on a non-magnetic substrate via an underlayer magnetic film, the underlayer magnetic film comprises a **multilayer** film of ≥ 3 layers of a magnetic film(s) and non-magnetic film(s) of an alloy based on Ti, Hf, Zr, or Mn, Si, Ge, semimetal based on Si or Ge, or a compd. selected from an oxide, carbide, boride, or nitride. Specifically, the compd. may comprise SiO₂, Al₂O₃, ZrO₂, MgO, CaO, and the magnetic film may comprise Co-Nb-X (X=Zr, Ti, Hf, Mo, W), Ni-Co-Y (Y=Zr, Ti, Hf, Mo, W), or Fe-Si-Z (Z=B, Al). A high-speed and high-d. magnetic recording/reprodn. app. comprising the above medium is also described.

IT 7440-21-3, Silicon, uses 12674-90-7, Iron 98, silicon 2 (atomic) 37214-91-8, Germanium 25, silicon 75 (atomic) 37255-61-1, Iron 95, silicon 5 (atomic) 72048-89-6, Germanium 80, silicon 20 (atomic)
RL: DEV (Device component use); USES (Uses)
(**multilayer** film for underlayer of perpendicular magnetic recording medium and magnetic recording/reprodn. app.)

RN 7440-21-3 HCAPLUS

CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

RN 12674-90-7 HCAPLUS

CN Iron alloy, base, Fe 99, Si 1 (9CI) (CA INDEX NAME)

Component	Component	Component
Percent	Registry Number	
Fe	99	7439-89-6
Si	1	7440-21-3

RN 37214-91-8 HCAPLUS

CN Silicon alloy, base, Si 54, Ge 46 (9CI) (CA INDEX NAME)

Component	Component	Component
Percent	Registry Number	
Si	54	7440-21-3
Ge	46	7440-56-4

RN 37255-61-1 HCAPLUS

CN Iron alloy, base, Fe 97, Si 2.6 (9CI) (CA INDEX NAME)

Component	Component	Component
Percent	Registry Number	
Fe	97	7439-89-6
Si	2.6	7440-21-3

RN 72048-89-6 HCAPLUS

CN Germanium alloy, base, Ge 91, Si 8.8 (9CI) (CA INDEX NAME)

Component	Component	Component
Percent	Registry Number	
Ge	91	7440-56-4

Si 8.8 7440-21-3

IC ICM G11B005-667
 CC 77-8 (Magnetic Phenomena)
 ST magnetic recording app multilayer film underlayer
 IT Magnetic films
 Magnetic memory devices
 (multilayer film for underlayer of perpendicular
 magnetic recording medium and magnetic recording/reprodn. app.)
 IT Films
 (multilayer; multilayer film for underlayer
 of perpendicular magnetic recording medium and magnetic
 recording/reprodn. app.)
 IT 12446-11-6, Aluminum zirconium oxide Al₂ZrO₅
 RL: DEV (Device component use); USES (Uses)
 (Al₂O₃•ZrO₂; multilayer film for underlayer of
 perpendicular magnetic recording medium and magnetic
 recording/reprodn. app.)
 IT 104244-64-6, Calcium magnesium oxide CaMgO₂
 RL: DEV (Device component use); USES (Uses)
 (MgO•CaO; multilayer film for underlayer of
 perpendicular magnetic recording medium and magnetic
 recording/reprodn. app.)
 IT 409-21-2, Silicon carbide, uses 1305-78-8, Calcium oxide (CaO),
 uses 1309-48-4, Magnesium oxide (MgO), uses 1314-23-4, Zirconium
 oxide (ZrO₂), uses 1344-28-1, Alumina, uses 7439-96-5,
 Manganese, uses 7440-21-3, Silicon, uses 7440-32-6,
 Titanium, uses 7440-56-4, Germanium, uses 7440-58-6, Hafnium,
 uses 7440-67-7, Zirconium, uses 7631-86-9, Silica, uses
 10101-39-0 10101-52-7 11148-13-3, Iron 20, nickel 80 (atomic)
 12007-23-7, Hafnium diboride 12008-21-8, Lanthanum boride
 12033-89-5, Silicon nitride, uses 12045-63-5, Titanium diboride
 12045-64-6, Zirconium diboride 12069-32-8, Boron carbide (B₄C)
 12069-85-1, Hafnium carbide 12070-08-5, Titanium carbide
 12070-14-3, Zirconium carbide 12621-05-5, Chromium 5, germanium 95
 (atomic) 12674-90-7, Iron 98, silicon 2 (atomic)
 14504-95-1 25583-20-4, Titanium nitride 25658-42-8, Zirconium
 nitride 25817-87-2, Hafnium nitride 37214-91-8,
 Germanium 25, silicon 75 (atomic) 37255-61-1, Iron 95,
 silicon 5 (atomic) 37274-26-3, Iron 50, platinum 50 (atomic)
 53801-50-6, Yttrium boride 72048-89-6, Germanium 80,
 silicon 20 (atomic) 80579-44-8, Chromium 30, zirconium 70 (atomic)
 87931-85-9, Cobalt 90, hafnium 3, niobium 7 (atomic) 97793-35-6,
 Cerium boride 99150-20-6, Hafnium 45, zirconium 55 (atomic)
 106698-99-1, Titanium carbide nitride (TiC0.7N0.3) 121229-13-8,
 Iron 60, platinum 40 (atomic) 137670-59-8, Cobalt 90, niobium 5,
 zirconium 5 (atomic) 149570-43-4, Hafnium 30, titanium 70 (atomic)
 156356-99-9, Cobalt 55, platinum 45 (atomic) 177575-08-5,
 Manganese 90, zirconium 10 (atomic) 186888-91-5, Chromium 15,
 cobalt 70, platinum 15 (atomic) 208260-58-6, Chromium 20, cobalt
 70, platinum 10 (atomic) 208260-62-2, Chromium 16, cobalt 81,
 tantalum 3 (atomic) 223516-48-1, Manganese 3, silicon 97 (atomic)
 224314-61-8, Titanium zirconium boride (Ti0.8Zr0.2B2) 313519-07-2,
 Manganese 5, silicon 95 (atomic) 313519-08-3, Cobalt 91, hafnium
 3, niobium 6 (atomic) 313519-10-7, Chromium 19, cobalt 70,
 platinum 8, tantalum 3 (atomic) 313519-12-9, Cobalt 18, samarium
 82 (atomic) 313519-14-1, Boron 3, iron 20, neodymium 77 (atomic)
 313519-15-2, Cobalt 25, nickel 71, zirconium 4 (atomic)
 313519-17-4, Chromium 35, titanium 65 (atomic) 313519-19-6, Cobalt
 91, molybdenum 3, niobium 6 (atomic) 313519-21-0, Chromium 17,

cobalt 79, tantalum 3, yttrium 1 (atomic) 313519-23-2, Cobalt 82, niobium 5, titanium 3 (atomic) 313519-25-4, Cobalt 92, hafnium 3, niobium 5 (atomic) 313519-27-6, Cobalt 94, niobium 4, tungsten 2 (atomic) 313519-29-8, Cobalt 10, nickel 88, zirconium 2 (atomic) 313519-31-2, Cobalt 28, nickel 69, titanium 3 (atomic) 313519-33-4, Cobalt 5, hafnium 4, nickel 91 (atomic) 313519-35-6, Cobalt 7, molybdenum 3, nickel 90 (atomic) 313519-37-8, Cobalt 15, nickel 83, tungsten 2 (atomic) 313519-39-0, Boron 2, iron 95, silicon 3 (atomic) 313519-41-4, Aluminum 3, iron 93, silicon 4 (atomic) 313519-44-7, Chromium 20, cobalt 70, platinum 8, tantalum 2 (atomic)

RL: DEV (Device component use); USES (Uses)
(multilayer film for underlayer of perpendicular magnetic recording medium and magnetic recording/reprodn. app.)

L29 ANSWER 19 OF 34 HCAPLUS COPYRIGHT 2006 ACS on STN

ACCESSION NUMBER: 2000:737091 HCAPLUS

DOCUMENT NUMBER: 133:289974

TITLE: Channel design to reduce impact ionization in heterostructure field-effect transistors

INVENTOR(S): Boos, J. Brad; Yang, Ming-jey; Bennett, Brian R.; Park, Doewon; Kruppa, Walter

PATENT ASSIGNEE(S): United States Dept. of the Navy, USA

SOURCE: U.S., 9 pp.

CODEN: USXXAM

DOCUMENT TYPE: Patent

LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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US 6133593	A	20001017	US 1999-358649	199907 23

PRIORITY APPLN. INFO.:	US 1999-358649	199907 23
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AB Heterostructure field-effect transistors (HFETs) and other electronic devices are fabricated from semiconductor layers to have reduced impact ionization. On to a 1st barrier layer there is added a unique 2nd sub-channel layer having high quality transport properties for reducing impact ionization. A 3rd barrier layer having a controlled thickness to permit electrons to tunnel through the layer to the sub-channel layer is added as a spacer for the 4th main channel layer. A 5th multilayer composite barrier layer is added which has at least a barrier layer in contact with the 4th channel layer and on top a 6th cap layer is applied. The device is completed by adding two ohmic contacts in a spaced apart relation on the 6th cap layer with a Schottky gate between them which is formed in contact with the 5th barrier layer. The 2nd sub-channel layer and the 4th main channel layers are made of materials which have the proper resp. energy gaps and ground state energies such that during use the transfer of hot electrons from the main channel into the sub-channel is made probable to reduce impact ionization in the main channel. In the preferred AlSb/InAs-based HFETs, the use of an InAs sub-channel layer under the main InAs channel improves the performance of the HEMTs particularly for gate

lengths in the deep-submicron regime. The devices exhibit higher transconductance, lower output conductance, reduced gate leakage current, higher operating drain voltage, and improved frequency performance.

IT 7440-21-3, Silicon, processes
 RL: DEV (Device component use); MOA (Modifier or additive use); PEP (Physical, engineering or chemical process); PROC (Process); USES (Uses)
 (in fabrication of heterostructure field-effect transistors)
 RN 7440-21-3 HCAPLUS
 CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

IT 11129-80-9, Platinum silicide 11148-21-3
 12738-91-9, Titanium silicide
 RL: DEV (Device component use); PEP (Physical, engineering or chemical process); PROC (Process); USES (Uses)
 (in fabrication of heterostructure field-effect transistors)
 RN 11129-80-9 HCAPLUS
 CN Platinum silicide (9CI) (CA INDEX NAME)

Component	Ratio	Component Registry Number
Si	x	7440-21-3
Pt	x	7440-06-4

RN 11148-21-3 HCAPLUS
 CN Germanium alloy, nonbase, Ge,Si (9CI) (CA INDEX NAME)

Component	Component Registry Number
Ge	7440-56-4
Si	7440-21-3

RN 12738-91-9 HCAPLUS
 CN Titanium silicide (9CI) (CA INDEX NAME)

Component	Ratio	Component Registry Number
Ti	x	7440-32-6
Si	x	7440-21-3

IC ICM H01L029-778
 INCL 257194000
 CC 76-3 (Electric Phenomena)
 IT 7440-21-3, Silicon, processes
 RL: DEV (Device component use); MOA (Modifier or additive use); PEP (Physical, engineering or chemical process); PROC (Process); USES (Uses)
 (in fabrication of heterostructure field-effect transistors)
 IT 1303-00-0, Gallium arsenide (GaAs), processes 1303-11-3, Indium arsenide (InAs), processes 1312-41-0 7440-05-3, Palladium, processes 7440-06-4, Platinum, processes 7440-32-6, Titanium, processes 7440-33-7, Tungsten, processes 7440-47-3, Chromium,

processes 7440-57-5, Gold, processes 11129-80-9,
 Platinum silicide 11148-21-3 12064-03-8 12070-08-5,
 Titanium carbide 12738-91-9, Titanium silicide
 22398-80-7, Indium phosphide (InP), processes 25152-52-7
 25617-97-4, Gallium nitride (GaN) 37382-15-3, Aluminum gallium
 arsenide (Al_{0.1}Ga_{0.9}As) 51680-21-8, Aluminum antimony gallium
 arsenide ((Al_{1-x}Ga_x)(Sb,As)) 106070-23-9, Aluminum indium arsenide
 (Al_{0.1}In_{0.9}As) 106070-25-1, Gallium indium arsenide (Ga_{0.1}In_{0.9}As)
 106097-44-3, Aluminum gallium nitride (Al_{0.1}Ga_{0.9}N) 106604-03-9,
 Antimony indium phosphide (Sb_{0.1}InP_{0.9}-1) 113959-17-4, Aluminum
 antimony phosphide (Al(Sb,P)) 114103-97-8, Aluminum indium
 arsenide (Al_{0.6}In_{0.4}As) 115184-93-5 117944-21-5, Aluminum
 antimony indium arsenide ((Al_{1-x}In_x)(Sb,As))
 RL: DEV (Device component use); PEP (Physical, engineering or
 chemical process); PROC (Process); USES (Uses)
 (in fabrication of heterostructure field-effect transistors)

REFERENCE COUNT: 3 THERE ARE 3 CITED REFERENCES AVAILABLE FOR
 THIS RECORD. ALL CITATIONS AVAILABLE IN
 THE RE FORMAT

L29 ANSWER 20 OF 34 HCPLUS COPYRIGHT 2006 ACS on STN
 ACCESSION NUMBER: 2000:552996 HCPLUS
 DOCUMENT NUMBER: 133:171034
 TITLE: Complimentary MIS semiconductor device and its
 fabrication
 INVENTOR(S): Mokami, Toru
 PATENT ASSIGNEE(S): NEC Corp., Japan
 SOURCE: Jpn. Kokai Tokkyo Koho, 16 pp.
 CODEN: JKXXAF
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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JP 2000223588	A2	20000811	JP 1999-26662	199902 03

PRIORITY APPLN. INFO.: JP 1999-26662 199902
03

AB A complimentary MIS semiconductor device comprises gate electrodes
 from a **multilayer** structure of conductor films having
 different work functions to give suitable threshold voltages: bottom
 conductor films having a thickness thin enough to change channel
 potentials. A method for fabricating the above device is also
 described.

IT 7440-21-3, Silicon, processes 11148-21-3
 12738-91-9, Titanium silicide
 RL: DEV (Device component use); PEP (Physical, engineering or
 chemical process); PROC (Process); USES (Uses)
 (**multilayer** gate electrode of complimentary MIS
 semiconductor device and its fabrication)

RN 7440-21-3 HCPLUS
 CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

RN 11148-21-3 HCPLUS
CN Germanium alloy, nonbase, Ge,Si (9CI) (CA INDEX NAME)

Component Component
 Registry Number

Ge 7440-56-4
Si 7440-21-3

RN 12738-91-9 HCAPLUS
CN Titanium silicide (9CI) (CA INDEX NAME)

Component	Ratio	Component Registry Number
Ti	x	7440-32-6
Si	x	7440-21-3

IC ICM H01L021-8238
ICS H01L027-092; H01L021-28; H01L029-43; H01L029-78
CC 76-3 (Electric Phenomena)
IT Films
Films
(elec. conductive; **multilayer** gate electrode of
complimentary MIS semiconductor device and its fabrication)
IT Electric conductors
Electric conductors
(films; **multilayer** gate electrode of complimentary MIS
semiconductor device and its fabrication)
IT Film electrodes
MIS devices
MISFET (transistors)
Semiconductor device fabrication
(**multilayer** gate electrode of complimentary MIS
semiconductor device and its fabrication)
IT 7439-88-5, Iridium, processes 7439-98-7, Molybdenum, process
7440-21-3, Silicon, processes 7440-33-7, Tungsten,
processes 7440-58-6, Hafnium, processes 7440-67-7, Zirconium
processes 11113-84-1, Ruthenium oxide 11148-21-3
12627-41-7, Tungsten silicide 12738-91-9, Titanium
silicide 25583-20-4, Titanium nitride
RL: DEV (Device component use); PEP (Physical, engineering or
chemical process); PROC (Process); USES (Uses)
(**multilayer** gate electrode of complimentary MIS
semiconductor device and its fabrication)

L29 ANSWER 21 OF 34 HCPLUS COPYRIGHT 2006 ACS on STN

ACCESSION NUMBER: 2000:461878 HCPLUS

DOCUMENT NUMBER: 133:273889

TITLE: Quantum well intersubband THz lasers and detectors

AUTHOR(S): Soref, Richard A.; Friedman, Lionel R.; Sun, Gregory; Noble, Michael J.; Ram-Mohan, L. R.

CORPORATE SOURCE: Sensors Directorate, Air Force Research Lab., AFRL/SHNC. Hanscom AFB. MA. USA

SOURCE: Proceedings of SPIE-The International Society for Optical Engineering (1999), 3795(Terahertz

and Gigahertz Photonics), 516-527

CODEN: PSISDG; ISSN: 0277-786X

PUBLISHER: SPIE-The International Society for Optical
EngineeringDOCUMENT TYPE: Journal
LANGUAGE: English

AB This paper presents modeling and simulation results on Si-based quantum-well intersubband THz detectors and THz lasers (tasers) in the 3 to 10 THz range where the intersubband transition energy is 12 to 41 meV. The incoherent cryogenically cooled (4 K to 20 K) quantum well terahertz detector (QWTD) consists of p-type Si_{0.9}Ge_{0.1} QWs with Si barriers on an Si substrate, or of p-Si_{0.85}Ge_{0.15}/Si on a relaxed Si_{0.97}Ge_{0.03} buffer on Si. The QWTD senses THz radiation at normal incidence (the XY polarization on the HH1 to LH1 transition) or at edge- illumination (the Z polarization on the HH1 to HH2 transition). Resonant-cavity enhancement, coupling to Si THz waveguides, and integration with SiGe transistor preamplifiers appear feasible for QWTDs. The quantum staircase taser is a simplified far-IR version of the quantum cascade laser in which each superlattice transfer region is replaced by a thin tunnel-barrier layer. The authors have adapted to Group IV the III-V idea of Sun, Lu, and Khurjin; the inverted mass taser. On a Si_{0.81}Ge_{0.19} substrate, an inverted effective mass exists in LH1 at kg is 0.013 Å⁻¹ in 9-nm single-wells of Si_{0.7}Ge_{0.3} with 5-nm Si barriers. Selective elec. injection of holes into LH1 at T is 77 K is postulated. This offers local-in-k-space LH1-HH1 population inversion and tasing at 7.2 THz. Since the taser emission is XY-polarized, the active MQW staircase (a set of identical square QWs) is suitable for insertion into a vertical cavity surface-emitting taser. The VCSET would have resonator thickness of $\lambda/2n$ is 6 μ m, and Bragg mirrors constructed from SiO₂/Si multilayers.

IT 7440-21-3, Silicon, properties 12017-12-8, Cobalt disilicide 12623-04-0, Germanium 30, silicon 70 (atomic)
37232-85-2, Germanium 15, silicon 85 (atomic)
51845-18-2, Germanium 10, silicon 90 (atomic)
119849-50-2, Germanium 97, silicon 3 (atomic)
124776-23-4, Germanium 3, silicon 97 (atomic)
RL: DEV (Device component use); PRP (Properties); USES (Uses)
(quantum well intersubband THz lasers and detectors)

RN 7440-21-3 HCAPLUS

CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

RN 12017-12-8 HCAPLUS

CN Cobalt silicide (CoSi₂) (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)

Si

$$\begin{array}{c} ||| \\ \text{Co} \equiv \text{Si} \end{array}$$

RN 12623-04-0 HCAPLUS

CN Germanium alloy, base, Ge 53, Si 47 (9CI) (CA INDEX NAME)

Component Component Component

Percent Registry Number

	Percent	Registry Number
Ge	53	7440-56-4
Si	47	7440-21-3

RN 37232-85-2 HCPLUS

CN Silicon alloy, base, Si 69,Ge 31 (9CI) (CA INDEX NAME)

Component Component Component

	Percent	Registry Number
Si	69	7440-21-3
Ge	31	7440-56-4

RN 51845-18-2 HCPLUS

CN Silicon alloy, base, Si 78,Ge 22 (9CI) (CA INDEX NAME)

Component Component Component

	Percent	Registry Number
Si	78	7440-21-3
Ge	22	7440-56-4

RN 119849-50-2 HCPLUS

CN Germanium alloy, base, Ge 99,Si 1.2 (9CI) (CA INDEX NAME)

Component Component Component

	Percent	Registry Number
Ge	99	7440-56-4
Si	1.2	7440-21-3

RN 124776-23-4 HCPLUS

CN Silicon alloy, base, Si 93,Ge 7.4 (9CI) (CA INDEX NAME)

Component Component Component

	Percent	Registry Number
Si	93	7440-21-3
Ge	7.4	7440-56-4

CC 73-11 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)

Section cross-reference(s): 76

IT 7440-21-3, Silicon, properties 7631-86-9, Silica,

properties 12017-12-8, Cobalt disilicide

12623-04-0, Germanium 30, silicon 70 (atomic)

37232-85-2, Germanium 15, silicon 85 (atomic)

51845-18-2, Germanium 10, silicon 90 (atomic)

119849-50-2, Germanium 97, silicon 3 (atomic)

124776-23-4, Germanium 3, silicon 97 (atomic)

RL: DEV (Device component use); PRP (Properties); USES (Uses)

(quantum well intersubband THz lasers and detectors)

REFERENCE COUNT: 26 THERE ARE 26 CITED REFERENCES AVAILABLE
FOR THIS RECORD. ALL CITATIONS AVAILABLE
IN THE RE FORMAT

L29 ANSWER 22 OF 34 HCPLUS COPYRIGHT 2006 ACS on STN

ACCESSION NUMBER: 1998:81123 HCPLUS

DOCUMENT NUMBER: 128:199639

TITLE: Thermoelectric materials by barrier/conductor
 multilayers
 INVENTOR(S): Nishimoto, Seiji
 PATENT ASSIGNEE(S): Honda Motor Co., Ltd., Japan
 SOURCE: Jpn. Kokai Tokkyo Koho, 5 pp.
 CODEN: JKXXAF
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 10032354	A2	19980203	JP 1996-185866	199607 16
JP 3526699	B2	20040517		
US 5886292	A	19990323	US 1997-895515	199707 16
PRIORITY APPLN. INFO.:			JP 1996-185866	A 199607 16

AB The title thermoelec. materials have an alternately laminated
 barrier/conductor **multilayer** by employing (a) 1st
 semiconductive conductor layers and (b) barrier layers which
 comprise (1) outer barrier layers contg. a 2nd semiconductive outer
 barrier layer and 1st/2nd semiconductive boundary layers and (2)
 inner barrier layers contg. a 2nd semiconductive outer barrier layer
 and 2 2nd semiconductive outer barrier layers. The thickness of the
 conductors (t1) and barriers (t2) has a relation as
 $t1 \leq t2 \leq 50t1$ so as to form a quantum well in each
 conductor layer.

IT 7440-21-3, Silicon, properties 11148-21-3
 12022-99-0, Iron silicide (FeSi₂)
 RL: DEV (Device component use); PEP (Physical, engineering or
 chemical process); PRP (Properties); PROC (Process); USES (Uses)
 (semiconductor layer for thermoelec. multilayer;
 thermoelec. materials by barrier/conductor multilayers)

RN 7440-21-3 HCAPLUS

CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

RN 11148-21-3 HCAPLUS
 CN Germanium alloy, nonbase, Ge,Si (9CI) (CA INDEX NAME)

Component	Component
Registry Number	
Ge	7440-56-4
Si	7440-21-3

RN 12022-99-0 HCAPLUS
 CN Iron silicide (FeSi₂) (6CI, 8CI, 9CI) (CA INDEX NAME)

Fe≡Si
|||
Si

IC ICM H01L035-14
ICS H01L035-16; H01L035-26
CC 76-6 (Electric Phenomena)
Section cross-reference(s): 56, 57
IT Molecular surface
(boundary layer, semiconductor; thermoelec. materials by
barrier/conductor **multilayers**)
IT Diffusion barrier
Quantum well heterojunctions
Thermoelectric materials
(thermoelec. materials by barrier/conductor **multilayers**)
)
IT Semiconductor devices
(thermoelec.; thermoelec. materials by barrier/conductor
multilayers)
IT 7440-21-3, Silicon, properties 11148-21-3
12022-99-0, Iron silicide (FeSi₂) 121515-47-7, Cobalt iron
silicide (Co0.1Fe0.9Si₂) 159679-85-3, Iron manganese silicide
(Fe0.9Mn0.1Si₂) 188921-84-8, Antimony germanium silver telluride
(Sb0.15Ge0.85Ag0.15Te1.15) 203799-91-1, Lead sodium selenide
telluride (PbNa0.01Se0.05Te0.95) 203799-92-2, Europium lead
telluride (Eu0.07Pb0.9Te)
RL: DEV (Device component use); PEP (Physical, engineering or
chemical process); PRP (Properties); PROC (Process); USES (Uses)
(semiconductor layer for thermoelec. **multilayer**;
thermoelec. materials by barrier/conductor **multilayers**)

L29 ANSWER 23 OF 34 HCAPLUS COPYRIGHT 2006 ACS on STN

ACCESSION NUMBER: 1998:79663 HCAPLUS
DOCUMENT NUMBER: 128:161932
TITLE: Thermoelectric materials
INVENTOR(S): Nishimoto, Kiyoji
PATENT ASSIGNEE(S): Honda Motor Co., Ltd., Japan
SOURCE: Jpn. Kokai Tokkyo Koho, 10 pp.
CODEN: JKXXAF
DOCUMENT TYPE: Patent
LANGUAGE: Japanese
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 10032355	A2	19980203	JP 1996-204167	199607 16
JP 3502724	B2	20040302		
US 5922988	A	19990713	US 1997-895378	199707 16
PRIORITY APPLN. INFO.:			JP 1996-204167	A
				199607 16

AB The material consists of an alternate multilayer from a conductive and a barrier semiconductor layer with formation of the interfaces between the layers to coarse surfaces having a ratio of the max. height of protrusions to the thickness of the barrier layer ≥ 0.1 . The material has good performance at an increased operating temp.

IT 7440-21-3, Silicon, properties

RL: PEP (Physical, engineering or chemical process); PRP (Properties); TEM (Technical or engineered material use); PROC (Process); USES (Uses)
(film; for barrier layers of thermoelec. alternate multilayers)

RN 7440-21-3 HCAPLUS

CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

IT 39300-22-6

RL: PEP (Physical, engineering or chemical process); PRP (Properties); TEM (Technical or engineered material use); PROC (Process); USES (Uses)
(film; for semiconductor conductive layers of thermoelec. alternate multilayers)

RN 39300-22-6 HCAPLUS

CN Silicon alloy, base, Si 80,Ge 20 (9CI) (CA INDEX NAME)

Component	Component	Component
Percent	Registry Number	
Si	80	7440-21-3
Ge	20	7440-56-4

IT 11148-21-3 12022-99-0, Iron silicide (FeSi₂)

RL: PEP (Physical, engineering or chemical process); PRP (Properties); TEM (Technical or engineered material use); PROC (Process); USES (Uses)
(film; for thermoelec. semiconductor conductor-barrier alternate multilayers)

RN 11148-21-3 HCAPLUS

CN Germanium alloy, nonbase, Ge, Si (9CI) (CA INDEX NAME)

Component	Component
Registry Number	
Ge	7440-56-4
Si	7440-21-3

RN 12022-99-0 HCAPLUS

CN Iron silicide (FeSi₂) (6CI, 8CI, 9CI) (CA INDEX NAME)

Fe \equiv Si

|||

Si

IC ICM H01L035-14

ICS H01L035-16; H01L035-26

CC 76-6 (Electric Phenomena)
 Section cross-reference(s): 56, 57

ST thermoelec alternate conductor barrier semiconductor
multilayer

IT Semiconductor materials
 (diffusion barriers; for formation of semiconductor
 conductor-barrier alternate **multilayer** thermoelec.
 materials)

IT Electric conductors
 Sputtering
 (for formation of semiconductor conductor-barrier alternate
multilayer thermoelec. materials)

IT Thermoelectric materials
 (semiconductor conductor-barrier alternate **multilayers**
 with coarse surface interfaces)

IT Diffusion barrier
 (semiconductor; for formation of semiconductor conductor-barrier
 alternate **multilayer** thermoelec. materials)

IT 7440-21-3, Silicon, properties
 RL: PEP (Physical, engineering or chemical process); PRP
 (Properties); TEM (Technical or engineered material use); PROC
 (Process); USES (Uses)
 (film; for barrier layers of thermoelec. alternate
multilayers)

IT 39300-22-6
 RL: PEP (Physical, engineering or chemical process); PRP
 (Properties); TEM (Technical or engineered material use); PROC
 (Process); USES (Uses)
 (film; for semiconductor conductive layers of thermoelec.
 alternate **multilayers**)

IT 1304-82-1, Bismuth telluride 11148-21-3 12022-99-0
 , Iron silicide (FeSi₂) 39280-96-1, Lead telluride
 RL: PEP (Physical, engineering or chemical process); PRP
 (Properties); TEM (Technical or engineered material use); PROC
 (Process); USES (Uses)
 (film; for thermoelec. semiconductor conductor-barrier alternate
multilayers)

L29 ANSWER 24 OF 34 HCPLUS COPYRIGHT 2006 ACS on STN
 ACCESSION NUMBER: 1998:79662 HCPLUS
 DOCUMENT NUMBER: 128:161931
 TITLE: Thermoelectric materials
 INVENTOR(S): Nishimoto, Kiyoji; Kitayama, Taku; Fujisawa,
 Yoshikazu
 PATENT ASSIGNEE(S): Honda Motor Co., Ltd., Japan
 SOURCE: Jpn. Kokai Tokkyo Koho, 5 pp.
 CODEN: JKXXAF

DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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JP 10032353	A2	19980203	JP 1996-185865	199607 16
JP 3497328	B2	20040216	JP 1996-185865	
PRIORITY APPLN. INFO.:				

199607
16

AB The material consists of an alternate multilayer from a conductive and a barrier semiconductor layer with interleaves of diffusion barrier layers there-between. The 1st semiconductor may be a FeSi2-, Si-Ge-, PbTe-, or BiTe-system substance, the 2nd semiconductor may be Si, or FeSi2-, Si-Ge-, or PbTe-system substance, and the barrier layer may be SiO, SiO2, TiO2, FeO, Fe2O3, SnO2, In2O3, CaMnO3, SiC, Si-B, or B4C.

IT 7440-21-3, Silicon, properties
 RL: DEV (Device component use); PRP (Properties); TEM (Technical or engineered material use); USES (Uses)
 (film; for semiconductor conductive layers in thermoelec.
 alternate multilayers)

RN 7440-21-3 HCPLUS

CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

IT 11148-21-3 12022-99-0, Iron silicide (FeSi2)
 39300-22-6
 RL: DEV (Device component use); PRP (Properties); TEM (Technical or engineered material use); USES (Uses)
 (film; for thermoelec. semiconductor conductor-barrier alternate multilayers)

RN 11148-21-3 HCPLUS

CN Germanium alloy, nonbase, Ge,Si (9CI) (CA INDEX NAME)

Component	Component
Registry Number	
Ge	7440-56-4
Si	7440-21-3

RN 12022-99-0 HCPLUS
 CN Iron silicide (FeSi2) (6CI, 8CI, 9CI) (CA INDEX NAME)

Fe≡Si
 |||
 Si

RN 39300-22-6 HCPLUS
 CN Silicon alloy, base, Si 80,Ge 20 (9CI) (CA INDEX NAME)

Component	Component	Component
Percent	Registry Number	
Si	80	7440-21-3
Ge	20	7440-56-4

IC ICM H01L035-14
 ICS H01L035-16; H01L035-26
 CC 76-6 (Electric Phenomena)
 Section cross-reference(s): 57
 ST thermoelec alternate multilayer diffusion barrier

IT Semiconductor materials
 RL: DEV (Device component use); PRP (Properties); USES (Uses)
 (diffusion barrier; for prepn. of semiconductor conductor-barrier
 alternate multilayer thermoelec. materials)

IT Diffusion barrier
 Sputtering
 (for prepn. of semiconductor conductor-barrier alternate multilayer thermoelec. materials)

IT Thermoelectric materials
 (semiconductor conductor-barrier alternate multilayers
 with diffusion barrier interleaves)

IT 409-21-2, Silicon carbide, properties 1309-37-1, Iron oxide
 (Fe₂O₃), properties 1312-43-2, Indium oxide (In₂O₃) 1345-25-1,
 Iron oxide (FeO), properties 7631-86-9, Silica, properties
 12069-32-8, Boron carbide (B₄C) 12177-86-5, Calcium manganese
 oxide (CaMnO₃) 12676-29-8, Boron silicate 13463-67-7, Titanium
 oxide (TiO₂), properties 18282-10-5, Tin oxide (SnO₂)
 113443-18-8, Silicon oxide (SiO)
 RL: DEV (Device component use); PRP (Properties); TEM (Technical or
 engineered material use); USES (Uses)
 (film; for diffusion barrier interleaves in thermoelec.
 semiconductor alternate multilayers)

IT 1304-82-1, Bismuth telluride 7440-21-3, Silicon,
 properties
 RL: DEV (Device component use); PRP (Properties); TEM (Technical or
 engineered material use); USES (Uses)
 (film; for semiconductor conductive layers in thermoelec.
 alternate multilayers)

IT 1314-91-6, Lead telluride (PbTe) 11148-21-3
 12022-99-0, Iron silicide (FeSi₂) 39300-22-6
 RL: DEV (Device component use); PRP (Properties); TEM (Technical or
 engineered material use); USES (Uses)
 (film; for thermoelec. semiconductor conductor-barrier alternate
 multilayers)

L29 ANSWER 25 OF 34 HCAPLUS COPYRIGHT 2006 ACS on STN
 ACCESSION NUMBER: 1995:986405 HCAPLUS
 DOCUMENT NUMBER: 124:19737
 TITLE: Forming a thin film for a multilayer
 semiconductor device
 INVENTOR(S): Agnello, Paul David; Cabral, Cyril, Jr.;
 Clevenger, Lawrence Alfred; Copel, Matthew
 Warren; D. Heurle, Francois Max; D'Heurle,
 Francois Max
 PATENT ASSIGNEE(S): International Business Machines Corp., USA
 SOURCE: Eur. Pat. Appl., 14 pp.
 CODEN: EPXXDW
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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EP 677868	A1	19951018	EP 1995-103863	199503 16
R: DE, FR, GB				
US 5624869	A	19970429	US 1994-226923	

KR 156064	B1	19981201	KR 1995-8351	199404 13
JP 08045875	A2	19960216	JP 1995-87195	199504 11
JP 3393731 US 5608266	B2 A	20030407 19970304	US 1995-458977	199504 12
PRIORITY APPLN. INFO.:				199506 02
US 1994-226923				A 199404 13

AB A method for stabilizing Co silicide/single-crystal Si, amorphous Si, polycryst. Si, germanide/cryst. Ge, polycryst. Ge structures or other semiconductor material structures is described, so that high-temp. processing steps (>750°) do not degrade the structural quality of the Co silicide/Si structure. The steps of the method include forming a silicide or germanide by either reacting Co with the substrate material and/or codepositing the silicide or germanide on a substrate, adding a selective element, either Pt or N, to the Co, and forming the silicide germanide by a std. annealing treatment. Alternatively, the Co alloy can be formed after the formation of the silicide or germanide, resp. As a result, the upper limit of the annealing temp. at which the silicide or germanide will structurally degrade is increased.

IT 7440-21-3, Silicon, processes 12017-12-8, Cobalt silicide (CoSi₂) 12727-59-2, Germanium 0-100, silicon 0-100 (atomic)

RL: DEV (Device component use); PEP (Physical, engineering or chemical process); PROC (Process); USES (Uses)
(forming thin films for multilayer semiconductor devices contg.)

RN 7440-21-3 HCAPLUS

CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

RN 12017-12-8 HCAPLUS

CN Cobalt silicide (CoSi₂) (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)

Si

|||
Co \equiv Si

RN 12727-59-2 HCAPLUS

CN Germanium alloy, base, Ge 0-100, Si 0-100 (9CI) (CA INDEX NAME)

Component	Component	Component
Percent	Registry Number	
Ge	0 - 100	7440-56-4
Si	0 - 100	7440-21-3

IC ICM H01L021-285
 CC 76-3 (Electric Phenomena)
 ST film formation **multilayer** semiconductor device
 IT Silicides
 RL: DEV (Device component use); PEP (Physical, engineering or
 chemical process); PROC (Process); USES (Uses)
 (forming thin films for **multilayer** semiconductor
 devices contg.)
 IT Annealing
 (in forming thin films for **multilayer** semiconductor
 devices)
 IT Semiconductor devices
 (**multilayer**, forming thin films for)
 IT 7440-21-3, Silicon, processes 7440-56-4, Germanium,
 processes 7440-56-4D, Germanium, compds. 7727-37-9, Nitrogen,
 processes 12017-12-8, Cobalt silicide (CoSi₂)
 12727-59-2, Germanium 0-100, silicon 0-100 (atomic)
 171499-21-1
 RL: DEV (Device component use); PEP (Physical, engineering or
 chemical process); PROC (Process); USES (Uses)
 (forming thin films for **multilayer** semiconductor
 devices contg.)

L29 ANSWER 26 OF 34 HCPLUS COPYRIGHT 2006 ACS on STN
 ACCESSION NUMBER: 1995:329869 HCPLUS
 DOCUMENT NUMBER: 122:203869
 TITLE: Solid state reaction of Co and Ti with
 epitaxially-grown Si_{1-x}Ge_x film on Si
 (100) substrate
 AUTHOR(S): Qi, Wen-Jie; Li, Bing-Zong; Huang, Wei-Ning; Gu,
 Zhi-Guang; Lu, Hong-Qiang; Zhang, Xiang-Jiu;
 Zhang, Ming; Dong, Guo-Sheng; Miller, David C.;
 et al.
 CORPORATE SOURCE: Department of Electronic Engineering, Fundan
 Univ., Shanghai, 200433, Peop. Rep. China
 SOURCE: Journal of Applied Physics (1995), 77(3),
 1086-92
 CODEN: JAPIAU; ISSN: 0021-8979
 PUBLISHER: American Institute of Physics
 DOCUMENT TYPE: Journal
 LANGUAGE: English
 AB The solid state reaction of Co and Ti with an epitaxially grown
 Si_{1-x}Ge_x strained layer was studied. The reaction was performed in
 a rapid thermal annealing system. The resulting films were
 characterized by Rutherford backscattering, Auger electron
 spectroscopy, XPS, x-ray diffractometry, and SEM. The elec.
 resistivity and Hall effect were measured at 77-300 K. Rapid
 thermal annealing of Co/Si_{0.8}Ge_{0.2} at 650° results in a
 Co(Si_{0.9}Ge_{0.1}) film with cubic cryst. structure. At higher temp.
 CoSi₂ is formed with Ge segregation towards the surface. After a
 multi-step annealing, a highly oriented CoSi₂ layer can be grown.
 For TiN/Ti/Si-Ge, the ternary phase of Ti(Si_{1-y}Ge_y)₂ is formed, with
 a smooth surface and with resistivity comparable to the lowest value
 exhibited by TiSi₂. The Co/Ti/Si-Ge/Si reaction was studied for the
 1st time, demonstrating that the uniformity of Co/Si-Ge reaction is
 improved by applying the Co/Ti bilayer. A TiN(O)/CoSi₂(Ge)/Si
multilayer structure is formed, and the CoSi₂(Ge) layer
 exhibits a strongly textured structure. Low temp. measurement
 reveals that the CoSi₂(Ge) layer has a resistivity slightly higher

than that of CoSi₂.

IT 12017-12-8, Cobalt silicide (CoSi₂)
 RL: FMU (Formation, unclassified); PRP (Properties); FORM
 (Formation, nonpreparative)
 (solid-state reaction of Co and Ti with epitaxial Si_{1-x}Ge_x film
 on Si(100) substrate)

RN 12017-12-8 HCAPLUS

CN Cobalt silicide (CoSi₂) (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)



IT 37380-03-3, Germanium 20, silicon 80 (atomic)
 RL: RCT (Reactant); RACT (Reactant or reagent)
 (solid-state reaction of Co and Ti with epitaxial Si_{1-x}Ge_x film
 on Si(100) substrate)

RN 37380-03-3 HCAPLUS

CN Silicon alloy, base, Si 61,Ge 39 (9CI) (CA INDEX NAME)

Component	Component Percent	Component Registry Number
Si	61	7440-21-3
Ge	39	7440-56-4

CC 78-9 (Inorganic Chemicals and Reactions)
 ST reaction cobalt titanium germanium silicon
 film
 IT Annealing
 (solid-state reaction of Co and Ti with epitaxial Si_{1-x}Ge_x film
 on Si(100) substrate)

IT 25583-20-4, Titanium mononitride
 RL: NUU (Other use, unclassified); USES (Uses)
 (capping layer; solid-state reaction of Co and Ti with epitaxial
 Si_{1-x}Ge_x film on Si(100) substrate)

IT 125135-18-4, Titanium germanide silicide (TiGe_{0.2}Si_{0.2})
 RL: FMU (Formation, unclassified); FORM (Formation, nonpreparative)
 (solid-state reaction of Co and Ti with epitaxial Si_{1-x}Ge_x film
 on Si(100) substrate)

IT 12017-12-8, Cobalt silicide (CoSi₂) 161582-66-7, Cobalt
 germanium silicide (CoGe_{0.1}Si_{0.9})
 RL: FMU (Formation, unclassified); PRP (Properties); FORM
 (Formation, nonpreparative)
 (solid-state reaction of Co and Ti with epitaxial Si_{1-x}Ge_x film
 on Si(100) substrate)

IT 7440-32-6, Titanium, reactions 7440-48-4, Cobalt, reactions
 37380-03-3, Germanium 20, silicon 80 (atomic)
 RL: RCT (Reactant); RACT (Reactant or reagent)
 (solid-state reaction of Co and Ti with epitaxial Si_{1-x}Ge_x film
 on Si(100) substrate)

L29 ANSWER 27 OF 34 HCAPLUS COPYRIGHT 2006 ACS on STN
 ACCESSION NUMBER: 1994:668945 HCAPLUS
 DOCUMENT NUMBER: 121:268945
 TITLE: Selective epitaxial silicon and selective
 titanium silicide in an industrial integrated
 cluster tool

AUTHOR(S): Regolini, J. L.; Margail, J.; Morin, C.;
 Gouy-Pailler, P.
 CORPORATE SOURCE: France Telecom, CNET/CNS, Meylan, Fr.
 SOURCE: Materials Research Society Symposium Proceedings
 (1994), 342(Rapid Thermal and Integrated
 Processing III), 249-54
 CODEN: MRSPDH; ISSN: 0272-9172

DOCUMENT TYPE: Journal
 LANGUAGE: English

AB Using an industrial integrated cluster reactor the authors have obtained selective epitaxial Si and selective TiSi₂ deposition. This is a 200 mm reactor in which epitaxial Si was obtained with <1% (1σ) thickness uniformity and <2% over a 25 wafer batch. Full selectivity of Si on oxide was obtained below a 20 torr working pressure using the DCS/H₂ gas system. No loading effect was detected. The main characteristics of this system are described with the most relevant results like:.. Sharp interfaces obtained in Si_{0.7}Ge_{0.3}/Si multilayer structures grown at 650°, abruptly doped epitaxial layers and residual defect d. TiSi₂ was selectively obtained with min. substrate consumption using the (H₂SiH₄ or DCS)/TiCl₄ chem. The elevated source and drain also was successfully tested by selective Si epitaxy followed by in situ selective TiSi₂ deposition to compensate for substrate consumption.

IT 7440-21-3P, Silicon, processes 12039-83-7P,
 Titanium silicide (TiSi₂) 12623-04-0P, Germanium 30,
 silicon 70 (atomic)
 RL: PEP (Physical, engineering or chemical process); SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); PROC (Process); USES (Uses)
 (selective epitaxy of Si, Ge-Si and TiSi₂ in industrial integrated cluster CVD reactor)

RN 7440-21-3 HCAPLUS

CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

RN 12039-83-7 HCAPLUS
 CN Titanium silicide (TiSi₂) (6CI, 8CI, 9CI) (CA INDEX NAME)

Si₂
 Ti₂Si₃

RN 12623-04-0 HCAPLUS
 CN Germanium alloy, base, Ge 53, Si 47 (9CI) (CA INDEX NAME)

Component	Component	Component
Percent	Registry Number	
Ge	53	7440-56-4
Si	47	7440-21-3

CC 76-3 (Electric Phenomena)
 Section cross-reference(s): 75
 IT 7440-21-3P, Silicon, processes 12039-83-7P,
 Titanium silicide (TiSi₂) 12623-04-0P, Germanium 30,
 silicon 70 (atomic)

RL: PEP (Physical, engineering or chemical process); SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); PROC (Process); USES (Uses)
 (selective epitaxy of Si, Ge-Si and TiSi₂ in industrial integrated cluster CVD reactor)

L29 ANSWER 28 OF 34 HCPLUS COPYRIGHT 2006 ACS on STN
 ACCESSION NUMBER: 1993:91538 HCPLUS
 DOCUMENT NUMBER: 118:91538
 TITLE: Production of a multilayer system and systems thus produced
 INVENTOR(S): Mantl, Siegfried; Bay, Helge
 PATENT ASSIGNEE(S): Forschungszentrum Juelich GmbH, Germany
 SOURCE: Eur. Pat. Appl., 11 pp.
 CODEN: EPXXDW
 DOCUMENT TYPE: Patent
 LANGUAGE: German
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
EP 510555	A1	19921028	EP 1992-106724	199204 18
EP 510555 R: BE, FR, GB, IT, NL	B1	19950726		
DE 4113143	A1	19921029	DE 1991-4113143	199104 23
DE 4113143	C2	19940804		
US 5250147	A	19931005	US 1992-866012	199204 08
JP 06177032	A2	19940624	JP 1992-101428	199204 21
CA 2066847	AA	19921024	CA 1992-2066847	199204 22
PRIORITY APPLN. INFO.:			DE 1991-4113143	A 199104 23

AB Multilayer structures are formed by the epitaxial growth of at least a 1st layer followed by the formation of a transition region by simultaneously depositing the material making up the 1st layer and at least a component of the 2nd layer and then subjecting the resulting structure to conditions under which a diffusive sepns. of phases occurs within the transition region to produce the multilayer structure.

IT 7440-21-3, Silicon, uses 12017-12-8, Cobalt disilicide

RL: USES (Uses)
 (deposition and diffusion of, in multilayer structure formation)

RN 7440-21-3 HCPLUS

CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

RN 12017-12-8 HCPLUS
 CN Cobalt silicide (CoSi₂) (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)



IT 37380-03-3
 RL: PROC (Process)
 (multilayer structures contg., manuf. of)
 RN 37380-03-3 HCPLUS
 CN Silicon alloy, base, Si 61,Ge 39 (9CI) (CA INDEX NAME)

Component	Component Percent	Component Registry Number
Si	61	7440-21-3
Ge	39	7440-56-4

IC ICM C30B033-02
 ICS H01L021-20
 CC 75-1 (Crystallography and Liquid Crystals)
 ST multilayer structure epitaxy diffusive sepn
 IT Diffusion
 Epitaxy
 (in multilayer structure formation)
 IT 1303-00-0, Gallium arsenide, uses 7440-21-3, Silicon, uses
 7631-86-9, Silica, uses 12017-12-8, Cobalt disilicide
 RL: USES (Uses)
 (deposition and diffusion of, in multilayer structure
 formation)
 IT 37380-03-3
 RL: PROC (Process)
 (multilayer structures contg., manuf. of)

L29 ANSWER 29 OF 34 HCPLUS COPYRIGHT 2006 ACS on STN
 ACCESSION NUMBER: 1992:581447 HCPLUS
 DOCUMENT NUMBER: 117:181447
 TITLE: IR longpass silicon filters
 INVENTOR(S): Fujii, Akito
 PATENT ASSIGNEE(S): Sumitomo Electric Industries, Ltd., Japan
 SOURCE: Jpn. Kokai Tokkyo Koho, 4 pp.
 CODEN: JKXXAF
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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JP 04136901	A2	19920511	JP 1990-259963	199009 28

PRIORITY APPLN. INFO.:

JP 1990-259963

199009
28

AB The filter comprises (1) a Si substrate, (2) a 1st metal fluoride, (3) a Ge or a Si-Ge alloy, and (4)-(5) a 2nd and a 3rd metal fluoride layer, wherein (2)-(5) form an antireflective/moisture-barrier laminate; (5) employs MgF₂ or CaF₂; the fluoride layers consist of cryst. micro-grains; and the linear transmittance is >90% in 3-5 μ m. The filter has a long-term stability in harsh environments.

IT 7440-21-3, Silicon, uses

RL: USES (Uses)
(antireflective film-coated IR longpass filters from, as
substrate)

BN 7440-31-3 HCARLIJS

CN Silicon (7CI 8CI 9CI) (CA INDEX NAME)

Si

IT 11148-21-3

RL: USES (Uses)
(antireflective multilayer coating from, for IR
longpass silicon filters)

RN 11148-21-3 HCAPLUS

CN Germanium alloy, nonbase, Ge,Si (9CI) (CA INDEX NAME)

Component Component
 Registry Number

Ge 7440-56-4
Si 7440-21-3

IC ICM G02B001-10

CC 73-12 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)

IT 7440-21-3. Silicon uses

RL: USES (Uses)
(antireflective film-coated IR longpass filters from, as
above.)

IT 7440-56-4, Germanium, uses 7783-40-6, Magnesium fluoride (MgF₂)
7789-75-5, Calcium fluoride (CaF₂), uses 11148-21-3
13709-38-1, Lanthanum fluoride (LaF₃) 13709-49-4, Yttrium fluoride

(YF3)
RL: USES (Uses)
 (antireflective multilayer coating from, for IR
 longpass silicon filters)

1-29 ANSWER 30 OF 34 HCABPLUS COPYRIGHT 2006 ACS OR STN

ACCESSION NUMBER: 1882:437763 HCABLIJS

ACCESSION NUMBER: 1992:4377
DOCUMENT NUMBER: 117:37763

DOCUMENT NUMBER: 117.37763
TITLE: Application of ion-beam-bevel sectioning to semiconducting and metallic layer structures

AUTHOR(S): Barkshire, I. R.; Roberts, R. H.; Greenwood, J. C.; Kenny, D. G.; Bruton, M.

CORPORATE SOURCE: C.; Kenny, P. G.; Prutton, M.
Dep. Phys., Univ. York, Heslington/York, YO1
5DD, UK

SOURCE: Institute of Physics Conference Series (1991),
 119(Electron Microsc. Anal. 1991), 17-20
 CODEN: IPCSEP; ISSN: 0951-3248

DOCUMENT TYPE: Journal
 LANGUAGE: English

AB Si/Si-Ge and Co/Co₂Si/CoSi/Si multilayer samples were bevel sectioned with 2 keV Ar⁺ and Xe⁺ ions. Subsequent imaging and line scanning in the York MULSAM instrument reveal that bevel angles as low as 1 mrad can be cut resulting in .apprx. 10 nm depth resoln. using a 200 nm diam. electron beam. Contributions of ion mixing to depth resoln. can be identified. Study of unexpected spectral features is possible after the bevel was cut.

IT 7440-21-3, Silicon, uses 11148-26-8, Germanium 14, silicon 86 (atomic) 12017-11-7, Cobalt monosilicide 12134-03-1, Cobalt silicide (Co₂Si)
 RL: USES (Uses)
 (ion beam sectioning of layers of)

RN 7440-21-3 HCAPLUS
 CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

RN 11148-26-8 HCAPLUS
 CN Silicon alloy, base, Si 70,Ge 30 (9CI) (CA INDEX NAME)

Component	Component	Component
Percent	Registry Number	
Si	70	7440-21-3
Ge	30	7440-56-4

RN 12017-11-7 HCAPLUS
 CN Cobalt silicide (CoSi) (6CI, 8CI, 9CI) (CA INDEX NAME)

$$\begin{array}{c} \equiv \\ \text{Si} \\ \text{Co} \end{array}$$

RN 12134-03-1 HCAPLUS
 CN Cobalt silicide (Co₂Si) (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)

Component	Ratio	Component
		Registry Number
Co	2	7440-48-4
Si	1	7440-21-3

CC 76-3 (Electric Phenomena)
 IT Cutting
 (ion-beam, of semiconductor structures and metal layers)
 IT Microscopy, electron
 (scanning, ion beam double sectioning of semiconductor structures and metal layers for studies by)
 IT 7440-21-3, Silicon, uses 7440-22-4, Silver, uses 7440-48-4, Cobalt, uses 7440-57-5, Gold, uses 11148-26-8, Germanium 14, silicon 86 (atomic) 12017-11-7, Cobalt

monosilicide 12134-03-1, Cobalt silicide (Co₂Si)
 RL: USES (Uses)
 (ion beam sectioning of layers of)

L29 ANSWER 31 OF 34 HCAPLUS COPYRIGHT 2006 ACS on STN
 ACCESSION NUMBER: 1990:110607 HCAPLUS
 DOCUMENT NUMBER: 112:110607
 TITLE: Magnetic multilayer films and
 manufacture thereof
 INVENTOR(S): Wakabayashi, Chizuko; Ishiwata, Nobuyuki;
 Matsumoto, Takayuki
 PATENT ASSIGNEE(S): NEC Home Electronics, Ltd., Japan
 SOURCE: Jpn. Kokai Tokkyo Koho, 9 pp.
 CODEN: JKXXAF
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 01119005	A2	19890511	JP 1987-276782	198710 31
PRIORITY APPLN. INFO.:			JP 1987-276782	198710 31

AB The film is made from a magnetic Fe-base film
 contg. Si, Ge, Y, or C (e.g., 1.5-3.5, 5-10,
 2-4, and <10 at.%, resp.) and a Permalloy film (e.g., 0.01-0.5 μ m
 and 1-10 nm thick, resp.). The multilayer film is formed
 on a nonmagnetic substrate and heat-treated (e.g., at
 500-800° in vacuum for 1 h). The multilayer has a
 high satn. magnetic flux d. and a low coercive force. The film is
 useful for magnetic head of high d. recording.

IT 11102-68-4 12715-55-8
 RL: PRP (Properties)
 (magnetic multilayer films from, with Permalloy films)
 RN 11102-68-4 HCAPLUS
 CN Iron alloy, base, Fe,Si (9CI) (CA INDEX NAME)

Component	Component
Registry Number	
Fe	7439-89-6
Si	7440-21-3

RN 12715-55-8 HCAPLUS
 CN Iron alloy, base, Fe 98,Si 1.8 (9CI) (CA INDEX NAME)

Component	Component	Component
Percent	Registry Number	
Fe	98	7439-89-6
Si	1.8	7440-21-3

IC ICM H01F010-14
 ICS G11B005-31

CC 77-8 (Magnetic Phenomena)
 ST iron base alloy Permalloy multilayer film
 IT Recording materials
 (magnetic, iron-base alloy-Permalloy multilayer films,
 for heads)
 IT 11102-68-4 12715-55-8 12716-37-9 39437-45-1
 51612-56-7
 RL: PRP (Properties)
 (magnetic multilayer films from, with Permalloy films)
 IT 11068-82-9, Permalloy
 RL: PRP (Properties)
 (magnetic multilayer films from, with iron-base alloy
 films)

L29 ANSWER 32 OF 34 HCAPLUS COPYRIGHT 2006 ACS on STN

ACCESSION NUMBER: 1987:506433 HCAPLUS
 DOCUMENT NUMBER: 107:106433
 TITLE: Optical recording medium
 INVENTOR(S): Okawa, Hideki
 PATENT ASSIGNEE(S): Toshiba Corp., Japan
 SOURCE: Jpn. Kokai Tokkyo Koho, 3 pp.
 CODEN: JKXXAF
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 62028940	A2	19870206	JP 1985-168005	198507 30
PRIORITY APPLN. INFO.:			JP 1985-168005	198507 30

AB The title optical recording medium, in which recording is effected
 by the optical property change induced by localized mixing or
 alloying of ≥2 recording layers, has a barrier interlayer
 formed by an anodic oxidn. of one of the recording layers. An
 optical disk was prep'd. by forming (1) an undercoat layer by CH4
 plasma polymn., (2) a metal recording layer
 (e.g., Al, Au, Pb, Sn, Te, Ni), (3) a barrier layer formed by anodic
 oxidn. of the metal layer, (4) a semiconductor
 recording layer (e.g., Ge, Si), and
 (5) an acetylcellulose overcoat layer. The barrier layer prevented
 the dispersion between the recording layers at room temp. so that
 the recording disk showed excellent storage stability before as well
 as after recording.

IT 7440-21-3, Silicon, uses and miscellaneous
 RL: TEM (Technical or engineered material use); USES (Uses)
 (optical recording material contg.)
 RN 7440-21-3 HCAPLUS
 CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

IC ICM G11B007-24
 ICS B41M005-26
 CC 74-12 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)
 ST optical recording medium multilayer; laser recording disk; anodic oxidn product barrier layer
 IT 7429-90-5, Aluminum, uses and miscellaneous 7439-92-1, Lead, uses and miscellaneous 7440-02-0, Nickel, uses and miscellaneous 7440-21-3, Silicon, uses and miscellaneous 7440-31-5, Tin, uses and miscellaneous 7440-56-4, Germanium, uses and miscellaneous 7440-57-5, Gold, uses and miscellaneous 13494-80-9, Tellurium, uses and miscellaneous
 RL: TEM (Technical or engineered material use); USES (Uses)
 (optical recording material contg.)

L29 ANSWER 33 OF 34 HCAPLUS COPYRIGHT 2006 ACS on STN

ACCESSION NUMBER: 1987:506430 HCAPLUS
 DOCUMENT NUMBER: 107:106430
 TITLE: Optical recording medium
 INVENTOR(S): Okawa, Hideki
 PATENT ASSIGNEE(S): Toshiba Corp., Japan
 SOURCE: Jpn. Kokai Tokkyo Koho, 4 pp.
 CODEN: JKXXAF
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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JP 62028943	A2	19870206	JP 1985-168015	198507 30
PRIORITY APPLN. INFO.:			JP 1985-168015	198507 30

AB The title optical recording medium, in which recording is effected by optical property change induced by localized mixing or alloying of ≥2 recording layers, has a barrier interlayer formed by plasma anodic oxidn. of one of the recording layers. An optical recording disk composed of (1) an undercoat layer, (2) a metal recording layer (e.g., Al, Au, Pb, Sn, Te), (3) a barrier layer formed by plasma anodic oxidn. of the metal layer surface, (4) a semiconductor recording layer (e.g., Ge, Si), and (5) an acetylcellulose overcoat layer was prep'd. The barrier layer prevented the diffusion between the recording layers at room temp. so that the recording disk showed excellent storage stability before as well as after recording.

IT 7440-21-3, Silicon, uses and miscellaneous
 RL: USES (Uses)
 (optical recording layer contg.)
 RN 7440-21-3 HCAPLUS
 CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

IC ICM G11B007-24
 ICS B41M005-26
 CC 74-12 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)
 ST optical recording medium multilayer; laser recording disk; plasma anodic oxidn barrier layer
 IT 7429-90-5, Aluminum, uses and miscellaneous 7439-92-1, Lead, uses and miscellaneous 7440-21-3, Silicon, uses and miscellaneous 7440-31-5, Tin, uses and miscellaneous 7440-56-4, Germanium, uses and miscellaneous 7440-57-5, Gold, uses and miscellaneous 13494-80-9, Tellurium, uses and miscellaneous
 RL: USES (Uses)
 (optical recording layer contg.)

L29 ANSWER 34 OF 34 HCAPLUS COPYRIGHT 2006 ACS on STN

ACCESSION NUMBER: 1987:415661 HCAPLUS
 DOCUMENT NUMBER: 107:15661
 TITLE: Optical recording medium
 INVENTOR(S): Okawa, Hideki
 PATENT ASSIGNEE(S): Toshiba Corp., Japan
 SOURCE: Jpn. Kokai Tokkyo Koho, 4 pp.
 CODEN: JKXXAF
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 62028939	A2	19870206	JP 1985-168004	198507 30
PRIORITY APPLN. INFO.:			JP 1985-168004	198507 30

AB The title optical recording medium, in which recording is effected by the optical property change induced by localized mixing or alloying of ≥ 2 recording layers, has a barrier interlayer formed by thermal oxidn. of one of the recording layer. An optical disk was composed of (1) an undercoat layer formed by CH₄-plasma polymn., (2) the metal recording layer (e.g., Al, Au, Pb, Sn, Te), (3) the barrier layer from the thermal oxidn. product formed by irradiating the surface of the metal recording layer with IR beam, (4) the semiconductor recording layer (e.g., Ge, Si), and (5) the acetylcellulose overcoat layer. The barrier layer prevented the dispersion between recording layers at room temp. so that the recording disk showed excellent storage stability before as well as after recording.

IT 7440-21-3, Silicon, uses and miscellaneous
 RL: TEM (Technical or engineered material use); USES (Uses)
 (optical recording medium contg.)

RN 7440-21-3 HCAPLUS
 CN Silicon (7CI, 8CI, 9CI) (CA INDEX NAME)

Si

IC ICM G11B007-24
ICS B41M005-26

CC 74-12 (Radiation Chemistry, Photochemistry, and Photographic and
Other Reprographic Processes)

ST optical recording medium multilayer; laser recording disk

IT 7429-90-5, Aluminum, uses and miscellaneous 7439-92-1, Lead, uses
and miscellaneous 7440-21-3, Silicon, uses and
miscellaneous 7440-31-5, Tin, uses and miscellaneous 7440-56-4,
Germanium, uses and miscellaneous 7440-57-5, Gold, uses and
miscellaneous 13494-80-9, Tellurium, uses and miscellaneous

RL: TEM (Technical or engineered material use); USES (Uses)
(optical recording medium contg.)

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